

SOLUBILITY CURVES

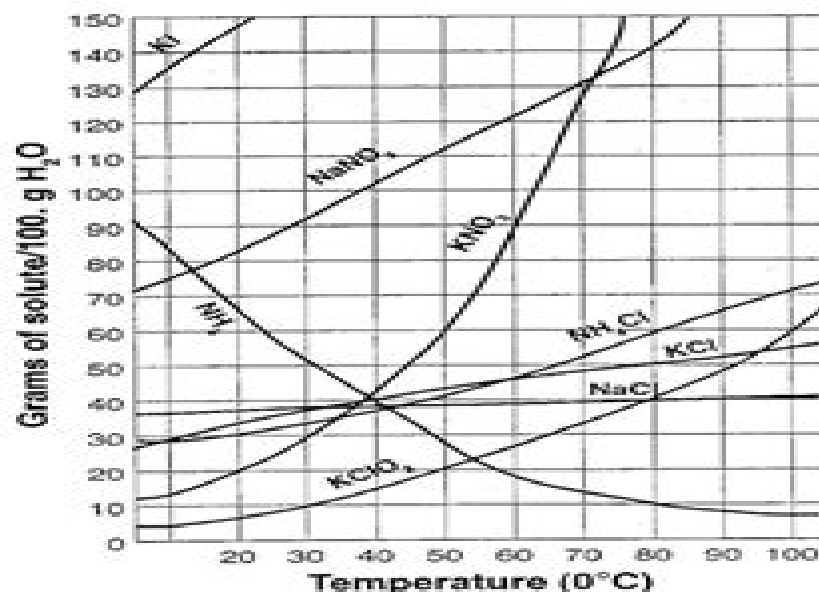
Name _____

Answer the following questions based on the solubility curve below.

1. Which salt is least soluble in water at 20° C? _____
2. How many grams of potassium chloride can be dissolved in 200 g of water at 80° C?

3. At 40° C, how much potassium nitrate can be dissolved in 300 g of water? _____
4. Which salt shows the least change in solubility from 0° – 100° C?

5. At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated?



6. A saturated solution of potassium chlorate is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? _____
7. What compound shows a decrease in solubility from 0° to 100° C? _____
8. Which salt is most soluble at 10° C? _____
9. Which salt is least soluble at 50° C? _____
10. Which salt is least soluble at 90° C? _____

Answer Key To Solubility Curve 1

SA Dillow



Answer Key To Solubility Curve 1:

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even in solid state Notwithstanding its inherently high complexity the crystallization process is part of our everyday lives from ice making in our homes to the most state of the art chemical and electronic industry In this book our purpose was to present new insights to the reader as well as crucial and very useful information for researchers working in this field while simultaneously creating a comprehensive text about crystallization processes which may serve as a starting point for people with different backgrounds

The Pearson Guide to Objective Chemistry for the AIEEE Singhal Atul,2010-09

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Mass Transfer Operations Mr. Rohit Manglik,2024-07-24 Discusses theoretical and practical aspects of mass transfer including distillation absorption and drying for students of chemical engineering

Handbook of Industrial Crystallization Allan Myerson,2002-01-08 Crystallization is an important separation and purification process used in industries ranging from bulk commodity chemicals to specialty chemicals and pharmaceuticals In recent years a number of environmental applications have also come to rely on crystallization in waste treatment and recycling processes The authors provide an introduction to the field of newcomers and a reference to those involved in the various aspects of industrial crystallization It is a complete volume covering all aspects of industrial crystallization including material related to both fundamentals and applications This new edition presents detailed material on crystallization of biomolecules precipitation impurity crystal interactions solubility and design Provides an ideal introduction for industrial crystallization newcomers Serves as a worthwhile reference to anyone involved in the field Covers all aspects of industrial crystallization in a single complete volume

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