Use your notebook to complete all of the following review questions. Please note that your completed notebook, corrected quizzes and unit review sheets are the <u>BEST</u> study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



- Complete the following metric conversions:

 - a) 75 g = __0.075__ kg e) 4.2 kL = _420 000_ cL
 - b) 4530 cm = 0.0453 km
- f) 83.2 cm = 832 mm g) 975 mg = _0.975_ g
- c) 0.8 A = 800 mA d) 3200 mA = __3,2_ A
 - h) 975 mg = 0.000975 kg
- Complete the following table using your knowledge of the Particle Theory of Matter:

STATE of MATTER	FORCES OF ATTRACTION	SPACING OF MOLECULES	MOVEMENT OF MOLECULES
solid	strongest	close together	least
liquid	weaker	farther apart	more
gas	weakest	farthest apart	most

- Changes of state are examples of _physical_ changes.
 - a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. EVAPOURATION 4, CONDENSATION



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change bubbles form precipitate forms light given off heat produced

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Padhraic Smyth

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- 1. Understanding the eBook Answers For Science Grade 9 Final Exam 2014
 - The Rise of Digital Reading Answers For Science Grade 9 Final Exam 2014
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Answers For Science Grade 9 Final Exam 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answers For Science Grade 9 Final Exam 2014
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Answers For Science Grade 9 Final Exam 2014
 - Personalized Recommendations
 - Answers For Science Grade 9 Final Exam 2014 User Reviews and Ratings

- Answers For Science Grade 9 Final Exam 2014 and Bestseller Lists
- 5. Accessing Answers For Science Grade 9 Final Exam 2014 Free and Paid eBooks
 - Answers For Science Grade 9 Final Exam 2014 Public Domain eBooks
 - Answers For Science Grade 9 Final Exam 2014 eBook Subscription Services
 - Answers For Science Grade 9 Final Exam 2014 Budget-Friendly Options
- 6. Navigating Answers For Science Grade 9 Final Exam 2014 eBook Formats
 - o ePub, PDF, MOBI, and More
 - Answers For Science Grade 9 Final Exam 2014 Compatibility with Devices
 - Answers For Science Grade 9 Final Exam 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answers For Science Grade 9 Final Exam 2014
 - Highlighting and Note-Taking Answers For Science Grade 9 Final Exam 2014
 - Interactive Elements Answers For Science Grade 9 Final Exam 2014
- 8. Staying Engaged with Answers For Science Grade 9 Final Exam 2014
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answers For Science Grade 9 Final Exam 2014
- 9. Balancing eBooks and Physical Books Answers For Science Grade 9 Final Exam 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answers For Science Grade 9 Final Exam 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answers For Science Grade 9 Final Exam 2014
 - Setting Reading Goals Answers For Science Grade 9 Final Exam 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answers For Science Grade 9 Final Exam 2014
 - Fact-Checking eBook Content of Answers For Science Grade 9 Final Exam 2014
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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