

◀ Homework Chapter 6

Exercise 6.39: Problems - The Mole Concept

▼ Part A

How many carbon atoms are in a diamond (pure carbon) with a mass of 75 mg?

Express your answer using two significant figures.



$N = 2.5 \cdot 10^{21}$

C atoms



You have already submitted this answer. Enter a new answer.

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Answers Mastering Chemistry Homework

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