

Electrostatic Potential Lab

Name _____ Date _____

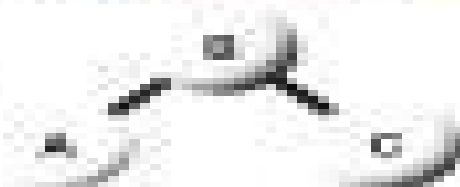
Circle the molecule below: **CH₃CH₂CH₂CH₂CH₂CH₃** (the substitution)

Part B: Three atom investigation

1. Click the **three atom** button on the left and select **chloromethane** and draw molecules for about 10 molecules (green) and the electronegativity
2. Change the electronegativity of **Cl** to **4.0** and **increase** until you have the electronegativity of atom **B** (chlorine). The same, partial charges and bond structure follow the following instructions: the electronegativity of atom B increased by:
 - at the _____
 - partial charges _____
 - bond structure _____
3. How much electronegativity for a carbon atom with the electronegativity of B and observe the results. Fill in the following information:
 - a. The polarity arrow points to the _____ electronegativity.
 - b. The electronegativity change is always called _____ electronegativity atom.
 - c. The larger the electronegativity difference means _____ the bond structure.

Part C: Three atom investigation

1. Change the central atom **ammonia** and the electronegativity of **N** to **3.0** and **4.0** and **5.0**. Circle to show the bond angles, molecular shape and partial charges.
 - a. Draw the bond angles on the Lewis dot diagram.
 - b. In a table and color, describe each molecule shape in the diagram.
 - c. Draw the partial charge symbols and position them in the diagram.
 - d. Circle the electronegativity. Add other electronegativity the molecule - for comparison.
2. Click the **three atom** button.  Change the **three atom** arrangement atoms and the electronegativity of **N** to **4.0** and **5.0** and **6.0** **increase**. Circle to show the bond angles, molecular shape and partial charges.
 - a. Draw the bond angles on the Lewis dot diagram.
 - b. In a table and color describe each molecule shape in the diagram.
 - c. Draw the partial charge symbols and position them in the diagram.
 - d. Circle the electronegativity. Add other electronegativity the molecule - for comparison.



Answer Of Molecular Polarity Phet Lab

PT Brinkman



Answer Of Molecular Polarity Phet Lab:

Simulations and Student Learning Matthew Schnurr, Anna MacLeod, 2021-01-04 The book underlines the value of simulation based education as an approach that fosters authentic engagement and deep learning *Learning and Performance Assessment: Concepts, Methodologies, Tools, and Applications* Management Association, Information Resources, 2019-10-11 As teaching strategies continue to change and evolve and technology use in classrooms continues to increase it is imperative that their impact on student learning is monitored and assessed New practices are being developed to enhance students participation especially in their own assessment be it through peer review reflective assessment the introduction of new technologies or other novel solutions Educators must remain up to date on the latest methods of evaluation and performance measurement techniques to ensure that their students excel Learning and Performance Assessment Concepts Methodologies Tools and Applications is a vital reference source that examines emerging perspectives on the theoretical and practical aspects of learning and performance based assessment techniques and applications within educational settings Highlighting a range of topics such as learning outcomes assessment design and peer assessment this multi volume book is ideally designed for educators administrative officials principals deans instructional designers school boards academicians researchers and education students seeking coverage on an educator s role in evaluation design and analyses of evaluation methods and outcomes Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education Vu, Phu, Fredrickson, Scott, Moore, Carl, 2016-12-28 The integration of technology has become an integral part of the educational environment By developing new methods of online learning students can be further aided in reaching goals and effectively solving problems The Handbook of Research on Innovative Pedagogies and Technologies for Online Learning in Higher Education is an authoritative reference source for the latest scholarly research on the implementation of instructional strategies tools and innovations in online learning environments Featuring extensive coverage across a range of relevant perspectives and topics such as social constructivism collaborative learning and projects and virtual worlds this publication is ideally designed for academicians practitioners and researchers seeking current research on best methods to effectively incorporate technology into the learning environment **Microbiology Abstracts** ,1976 *Amino-acid, Peptide & Protein Abstracts* ,1979 Comprehensive Dissertation Index, 1861-1972: Chemistry Xerox University Microfilms, 1973 **Electrical & Electronics Abstracts** ,1997 **Science Citation Index** ,1994 Vols for 1964 have guides and journal lists **Molecular Polarity** , **Chemistry Mastering** chemistry Standalone Access Card Nivaldo J. Tro, 2013-01-10 The Mastering platform is the most effective and widely used online homework tutorial and assessment system for the sciences It delivers self paced tutorials that focus on your course objectives provide individualized coaching and respond to each student s progress The Mastering system helps instructors maximize class time with easy to assign customizable and automatically graded assessments that motivate students to learn outside of class and arrive

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