

100) Carbon tetrachloride consists of one carbon and four chlorine atoms. How many chlorine atoms are present in 1.00 mole of CCl_4 molecules?

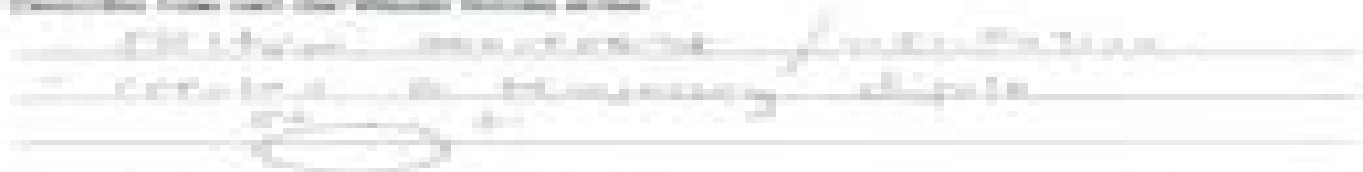
$$\text{one mole of } \text{CCl}_4 \text{ contains } 4 \text{ moles of } \text{Cl} \text{ atoms}$$

$$1 \text{ mole } \text{CCl}_4 \times \frac{4 \text{ moles Cl}}{1 \text{ mole } \text{CCl}_4} = 4 \text{ moles Cl}$$

$$\text{answer} = 4 \text{ moles Cl} \quad \text{100}$$

101) There are 4 monovalent atoms in each sulfur hexafluoride molecule.

102) Draw the Lewis structure for sulfur hexafluoride.



103) Draw the Lewis structure for sulfur hexafluoride.



104) Sulfur hexafluoride is a compound of sulfur and fluorine.

105) Draw the Lewis structure for sulfur hexafluoride.

What is the oxidation number of sulfur in SF_6 ?



Answer To F321 Chemistry May Series

John Anderson



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