

Atomic Probing Structure of the Atom

Name _____
Date _____



Purpose:

To determine the structure of the atom by using alpha particles and a lead box.

Part I: Finding Structure of the Atom by Using Alpha Particles

Alpha particles are positively charged particles with a mass of 4 atomic mass units. They are emitted from the nucleus of a radioactive substance. In this experiment, alpha particles are used to probe the structure of the atom. The results of the experiment are as follows:

Observation 1: Alpha particles are deflected by a magnetic field. This indicates that they have a positive charge. (Reference: [1])

Alpha particles are deflected by a magnetic field. This indicates that they have a positive charge. (Reference: [1])

Observation 2: Alpha particles are deflected by a lead box. This indicates that they have a mass. (Reference: [2])

Alpha particles are deflected by a lead box. This indicates that they have a mass. (Reference: [2])

Observation 3: Alpha particles are deflected by a lead box. This indicates that they have a mass. (Reference: [3])

Alpha particles are deflected by a lead box. This indicates that they have a mass. (Reference: [3])

Answers Lesson 11 Atomic Pudding

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Answers Lesson 11 Atomic Pudding:

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Atomic and Molecular Structure Walter J. Lehmann, 1972

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