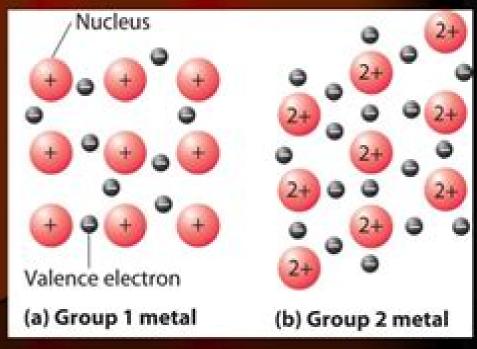
# Metallic bonding

- What happens when a metal bonds with another metal?
- Both metal atoms give up their valence electrons.

 This forms a "sea of electrons" that fills the space between the metal cations. The attraction of the positively charged metal cations to the negative valence electrons is what

holds the metal together.



# **Answers For Ionic Metallic Bonding**

**Mocktime Publication** 

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## **Answers For Ionic Metallic Bonding Introduction**

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