



All Breed Dog Guide

Bright George



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 $\text{NH}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{NH}_4\text{SO}_4$
 $\text{C}_5\text{H}_9\text{O}_2 + \text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{PbH}_3\text{PO}_4 + \text{H}_2\text{Pb}_3\text{PO}_4 + 2\text{Li}_3\text{N} + \text{NH}_4\text{NO}_3 + \text{LiNO}_3 + \text{NH}_4\text{N}_3$

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displacement 2 $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$ double displacement 3 $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$ combustion 4 $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$

single displacement 5 $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$ decomposition 6 $\text{P}_4 + 3\text{O}_2 \rightarrow \text{P}_2\text{O}_3$

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2 $2\text{Sr} + \text{C}_2\text{H}_2 + 3\text{O}_2 \rightarrow 2\text{SrCO}_3 + \text{H}_2\text{O}$ 3 $\text{LiBr} + \text{F}_2 \rightarrow \text{LiF} + \text{Br}_2$ 4 $2\text{Na} + \text{F}_2 \rightarrow 2\text{NaF}$ b balance the chemical reactions and identify each type

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double displacement 3 $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 3\text{MgO} + 2\text{Fe}$ type of reaction single displacement 4 $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$ type of reaction

combustion 5 $2\text{PbSO}_4 + 2\text{PbSO}_3 + 1\text{O}_2$ type of reaction

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