Name:	Date:	
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#### Student Exploration: Equilibrium and Concentration

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Vocabulary: chemical equilibrium, concentration, equilibrium, equilibrium constant, reaction quotient, reversible reaction

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

Gary has \$5,000 in his bank account and earns a modest salary. Every month he pays for rent, food, utilities, and entertainment.

- A. How will Gary's account change if he saves more than he spends? goes up
- B. How will Gary's account change if he spends more than he saves? goes down
- C. What happens if Gary spends exactly as much as he saves? stays teh same

#### Gizmo Warm-up

If Gary spends exactly as much as he earns, his savings will be in equilibrium. Equilibrium occurs when two opposing processes occur at the same rate, leading to no net change. In the Equilibrium and Concentration Gizmo, you will investigate how equilibrium can occur in chemical reactions.

To begin, check that Reaction 1 is selected. Set Moles  $NO_2$  to 8 and Moles  $N_2O_4$  to 0.



Click Play ( ) and observe the colliding molecules. What do you notice?

the molecules combine

In the Gizmo, a blue flash appears every time two reactants combine to form a product. A red flash appears every time a product dissociates into reactants.

Click Reset (2), and set Moles NO<sub>2</sub> to 0 and Moles N<sub>2</sub>O<sub>4</sub> to 8. Click Play.

What do you notice now? they break apart

 When a reaction can proceed in either direction, it is a reversible reaction. Based on what you have observed, is the synthesis of NO<sub>2</sub> into N<sub>2</sub>O<sub>4</sub> a reversible reaction? Explain.

yes

# **All Gizmo Answer Key**

**Alicia Cole, Lee Fanning** 



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