Reading the Periodic Table KEY

1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 3.

Na. Ma. AU SUP, S. CL Ar.

3. Name the elements in Period 2 that are metals.

Lat. Block

4. Name the elements in Period I that are nonmetals.

141, 140 et

- 5. Which element in Period 3 has the least mans? Sodium (No.)
- Which element in Period 4 has the most mass? Krypton (Kr)
- 7. To which group family do these belong?
 - a. Cobalt transition metals
 - b. Sodinan alkali metals.
 - c. Argon robble garners

- d. Silicon group 4/ metalloids
- e. Chlorine halogens
- Magnesium alkaline earth metals

- 8. Find from on the periodic table:
 - a. What is the atomic number of iron?
 - b. To which group does iron belong? Transition metals
- 9. Name the elements that have properties similar to that of chlorine.

fluorine, bronsine, indine, ustatine, tennessine

10. Fill in the chart below. The first now has been completed as an example.

Element	Symbol	Atomic Number	Pertiod	Group
sculture	No	11		7
emygomi -	0	8	- 2	1.6
krypton	Kr	346	-	- 18
gold	A/m	79	. 4	11
cademission	4.9	48	3	12
manganese	Min	23	4	9

Answer Key To The Periodic Table Homework

Laura Billings, Terry Roberts

Answer Key To The Periodic Table Homework:

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