

2. The solution of 2.25. (Homework 2.25, 2)

Problems 2.25(a), 2.25(b) and 2.25(c) require the following:

2.25(a). The volume of a spherical molecule can be approximated by a cubic volume. The side parameter of the molecule is the volume and V_0 is. Approximately, we can find the relationship by considering a spherical molecule of volume V_0 with radius $r_0 = (3V_0/4\pi)^{1/3}$. When water volume contained in the molecule that is partitioned for the water of the water molecule is estimated by the following, the volume V_0 can be approximated by the following (the volume of the water molecule is estimated by $V_0 = 30 \text{ Å}^3$). Calculate the value of r_0 (approximately) by using the value of V_0 (see the following).

The volume of the molecule is $V_0 = (4/3)\pi r_0^3 = (4/3)\pi r_0^3$ (see the following). Therefore,

for $r_0 = (3V_0/4\pi)^{1/3}$ and $V_0 = 30 \text{ Å}^3$, the value of r_0 is (see the following).

$$r_0 = \left(\frac{3V_0}{4\pi} \right)^{1/3} = \left(\frac{3 \times 30 \text{ Å}^3}{4\pi} \right)^{1/3} = \left(\frac{225}{4\pi} \right)^{1/3} \approx 2.1 \text{ Å}$$

2.25(b). Calculate the volume of the molecule of C_2H_6 from the volume of the molecule.

Therefore, the value for the volume of the molecule is (see the following).

$$V_0 = \frac{30}{4\pi} \left(\frac{4\pi}{3} \right)^{1/3} \left(\frac{30}{4\pi} \right)^{1/3} \approx 30 \text{ Å}^3$$

$$V_0 = \frac{30}{4\pi} \left(\frac{4\pi}{3} \right)^{1/3} \left(\frac{30}{4\pi} \right)^{1/3} \approx 30 \text{ Å}^3$$

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