



Name: _____

Date: _____

Chemistry – Study Guide

Ch 10 Study Guide Due: _____

Chapter 10 Test – _____

When studying for this test, use your 'do nows', notes, homework, class handouts, and your textbook.

Vocabulary

Avogadro's number

Mole

molar mass

equivalence statement

conversion factor

For all questions involving calculations, you must show all work, round your answers to the correct number of significant figures, and label your final answers with the correct units.

Measuring Matter - Chapter 10 – Section 10.1 – Pgs. 320-324

- Which of the following is NOT a representative particle?
 - Atom
 - Anion
 - Cation
 - All are representative particles
- Avogadro's number of representative particles is equal to one
 - Kilogram
 - Gram
 - Mole
 - Amu
- How many atoms of potassium does 1 mol of potassium contain?

$$6.02 \times 10^{23} \text{ atoms}$$

- How many moles of xenon are equivalent to 5.66×10^{26} atoms?

$$\frac{5.66 \times 10^{26} \text{ atoms}}{1} \times \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ atoms}} = 940. \text{ mol Xe}$$

Answers For Study Guide Mass Moles

N Noddings



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