Name \_\_\_\_\_

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# TYPES OF CHEMICAL REACTIONS

identify the type of chemical reaction (synthesis, decomposition, single displacement or replacement, double displacement or replacement).

1	$AB + C \rightarrow AC + B$	Single replacement
2	AB + CD AD + CB	Double replacement
3	$AB \rightarrow A + B$	Decomposition
4	$A + B \rightarrow AB$	Synthesis
5		Decomposition
6		Double replacement
7	○ • ○ ○ • ○ • ○	Single replacement
8	→	Synthesis
9	CaCO <sub>3</sub> → CaO + CO <sub>2</sub>	Decomposition
10	NaCl + AgNO <sub>3</sub> → NaNO <sub>2</sub> + AgCl	Double replacement
11	$Zn + 2HCl \rightarrow ZnCl_2 + H_2$	Single replacement
12	$Na(s) + Cl_2(g) \rightarrow 2NaCl(s)$	Synthesis
13	HCl + NaOH → NaCl + H <sub>2</sub> O	Double replacement
14	$2H_2O \rightarrow 2H_2 + O_2$	Decomposition
15	$Zn(s) + CuSO_a(aq) \rightarrow ZnSO_a(aq) + Cu(s)$	Single replacement

# **Answers For Types Of Reactions**

William C. Groutas, Athri D. Rathnayake

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while part B contains questions of somewhat greater difficulty suitable for students taking an honors course in organic chemistry or a beginning graduate course Detailed answers are provided to all questions so students can check their answers and important points are highlighted in each answer Special emphasis has been placed on the selection of questions to ensure that each question illustrates one or more fundamental principles of organic chemistry Interspersed throughout the book are minireviews that cover the material pertaining to a particular topic The specific literature references corresponding to each question are included and students can look up those references for more contextual information Includes a large number of carefully selected mechanism questions and step by step solutions including explanatory comments 

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