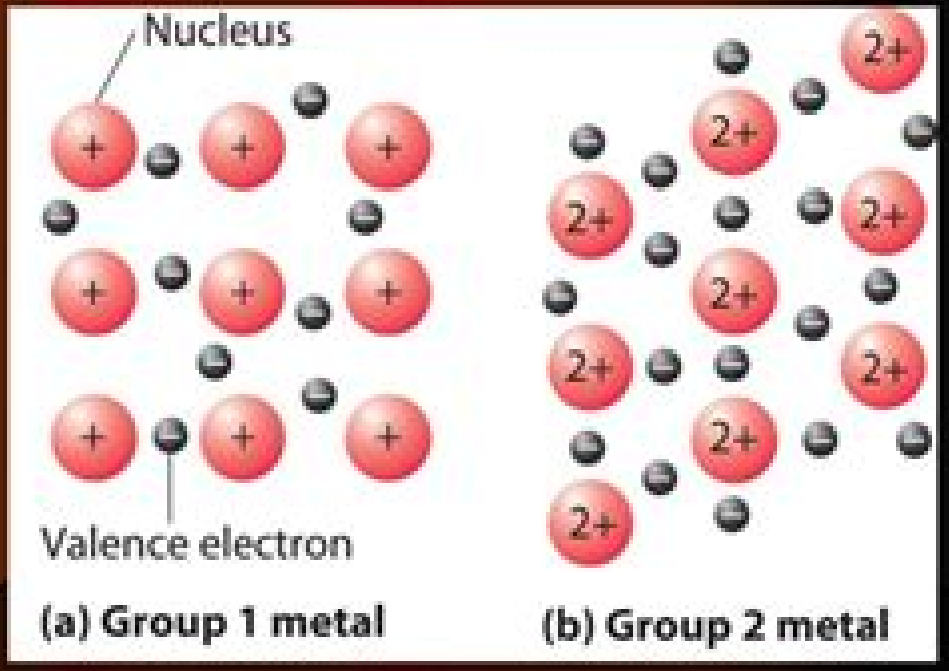
Metallic bonding

- What happens when a metal bonds with another metal?
- Both metal atoms give up their valence electrons.
- This forms a "sea of electrons" that fills the space between the metal cations. The attraction of the positively charged metal cations to the negative valence electrons is what holds the metal together.



Answers For Ionic Metallic Bonding

Ying Liu

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