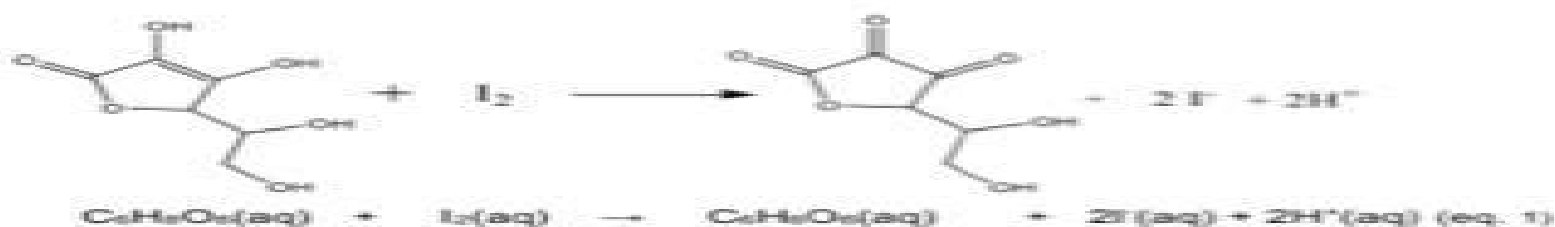


Volumetric Analysis for Vitamin C

The labelling and quantification of active ingredients in drugs is strictly controlled in Canada under the Food and Drugs Act. You can be fairly confident when you buy a prescription drug or over-the-counter medication that the listed ingredients are present in their indicated quantities. Vitamins and minerals are regulated as a sub-set of drugs and are governed by the Natural Health Products Regulation. While the regulations assure quality control, and require that companies report accurate and detailed information about the safety, quality, and efficacy of their products, the government does not have the resources to check quality control in the plethora of supplements on the market. During manufacture of large batches of pills, for example, problems may arise with ensuring a homogenous distribution of active ingredients. A recent study from the US (where regulations are more lax) found a wide variance in how much vitamin D was present in supplements compared to what was reported on the bottle.¹ A study such as this indicates the importance of analytical methods in determining how much of the active ingredient is truly present as compared to the label.

This experiment will quantitatively analyze an over-the-counter Vitamin C tablet for the mass of Vitamin C. Iodine reacts with Vitamin C by an oxidation process as indicated by the following reaction equation:



Note that one mole of iodine reacts with one mole of Vitamin C. In this experiment, a solution of Vitamin C (made from the tablet) is added to a solution containing excess iodine (with known concentration). Vitamin C is the limiting reactant; all of the Vitamin C reacts and leaves some iodine unreacted. The amount of iodine in excess is determined by titration with the standardized sodium thiosulfate from Lab D. Since the total (initial) amount of iodine was known, and the amount of excess iodine was determined in the titration, the amount of iodine that reacted with vitamin C can be determined by the difference. Thus, the amount of vitamin C can be determined.

The process of titrating an excess amount of a reactant not consumed in a reaction is called a 'back titration'. The related equations from Lab D are reproduced here for reference:



1. Leblanc, E.S.; Perrin, N.; Johnson, J.D. Over-the-Counter and Compounded Vitamin D: Is Potency What We Expect? *JAMA Intern Med.* **2013**, *173*, 585-586.

Analysis Of Vitamin C Lab Answers

R Sanford



Analysis Of Vitamin C Lab Answers:

Food Analysis Laboratory Manual S. Suzanne Nielsen, 2010-03-20 This second edition laboratory manual was written to accompany Food Analysis Fourth Edition ISBN 978 1 4419 1477 4 by the same author The 21 laboratory exercises in the manual cover 20 of the 32 chapters in the textbook Many of the laboratory exercises have multiple sections to cover several methods of analysis for a particular food component of characteristic Most of the laboratory exercises include the following introduction reading assignment objective principle of method chemicals reagents precautions and waste disposal supplies equipment procedure data and calculations questions and references This laboratory manual is ideal for the laboratory portion of undergraduate courses in food analysis *General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems *Chemistry Education in the ICT Age* Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings **Safety-Scale Lab Exp Biochem 2e** Spencer L. Seager, Michael R. Slabaugh, 1994-05 **Clinical Molecular Diagnostics** Shiyang Pan, Jinhai Tang, 2021-07-08 This book covers the discovery of molecular biomarkers the development of laboratory testing techniques and their clinical applications focusing on basic research to clinical practice It introduces new and crucial knowledge and ethics of clinical molecular diagnosis This book emphasizes the applications of clinical molecular diagnostic test on health management especially from different diseased organs It lets readers to understand and realize precision healthcare

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Automated urine chemistry by dipstrip and automated urinary sediment microscopy are discussed in some detail as this technology increasingly will be incorporated in veterinary laboratory practice The text is accompanied by hundreds of high quality photographs and medical illustrations that highlight common and rare findings from the urine of dogs and cats Numerous algorithms suggest possible pathways for the diagnosis and treatment of urinary disorders This book is destined to be widely used in veterinary hospitals by seasoned and young attending veterinarians seeking to know more about urinalysis as well as by veterinary laboratory technicians Several sections of this book will be useful in the teaching of basic concepts to veterinary students The authors have also included An introduction to the philosophy of urinalysis and the detailed information about collection of urine from dogs and cats Comprehensive explorations of urine sample handling preparation and analysis Detailed aspects of USG as the most important physical property of canine and feline urine In depth discussions of urine chemistry dipstrip evaluation for pH protein occult blood glucose ketones and bilirubin Extensive detailing of urinary sediment microscopy Sections on FAQ and urinalysis case examples allow the reader to test their knowledge about urinalysis Urinalysis in the Dog and Cat is an essential reference for primary care veterinarians veterinary technicians veterinary students those in advanced training programs and specialists interested in learning more about disorders of the urinary tract

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status and how attention to nutrition can help prevent illness or speed recovery describes the impact of severe stresses on nutrition status describes ways of feeding clients who cannot eat conventional foods and examines the specific dietary care required for clients with particular medical conditions *Nutrition Abstracts and Reviews* ,2006 Annual Report of the Minister of Agriculture Ontario. Department of Agriculture,1960 *Government Reports Announcements* ,1974-03

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