

Name: _____

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Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 3.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 3 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

Fluorine, bromine, iodine, astatine, tennessine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
sodium	Na	11	3	1
oxygen	O	8	2	16
krypton	Kr	36	4	18
gold	Au	79	6	11
cadmium	Cd	48	5	12
manganese	Mn	25	4	7

Answer Key To The Periodic Table Homework

Pam Walker, Elaine Wood



Answer Key To The Periodic Table Homework:

Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket

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