Name:	
Date:	Period:

Chemistry: The Periodic Table

Directions: Answer each of the following questions. You need not use complete sentences.

What is a period? How many are there in the periodic table?

A period is a row of elements on the periodic table, in which the elements are ordered one proton more each place from left to right. A period also often helps to represent an electron energy level and the order in which electrons are filled into their sublevels. There are 7 periods.

What is a group (also called a family)? How many are there in the periodic table?

Groups are the columns that are found top to bottom in the table. They show elements that have the same sublevel and number of electrons in their outer shell, valence electrons. Often elements in a group share any similar chemical properties because of this. There are 18 groups in the periodic table.

State the number of valence electrons in an atom of:

a. sulfur b. calcium c. chlorine d.arsenic 7 5

Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10 b. 18 c. 36 d. 90

Neon, Ne Argon, Ar Krypton, Kr Thorium, Th

List, by number, both the period and group of each of these elements.

	Symbol	Period	Group
a. beryllium	Be	2	2
b. Iron	Fe	4	-8
c. lead	Pb	6	1.4

Which of the following pairs of elements belong to the same period?

a. Na and Cl b. Na and Li c. Na and Cu d. Na and Ne

Which of the following pairs of elements belong to the same group?

a. H and He b. Li and Be c. C and Pb d. Ga and Ge

Would you expect strontium to be, chemically, more similar to calcium or rubidium and WHY?

Calcium, because they share a common amount of valence electrons. Valence electrons are the main contributor to chemical properties and how elements interact with other matter. Both strontium and calcium have two electron on the the "s" sublevel, 5s² and 4s² What are the Group 1 elements called?

Answers Key To Chemistry Radioactivity Packet 15

Joacim Rocklöv

Answers Key To Chemistry Radioactivity Packet 15:

High Level Radioactive Waste Management ,1995 Posture Hearings (NASA and FAA) United States. Congress. House. Committee on Science and Technology,1980 Radioactive Waste Management ,1981

Environmental Chemistry in the Lab Ruth Ann Murphy,2022-08-31 Environmental Chemistry in the Lab presents a comprehensive approach to modern environmental chemistry laboratory instruction together with a complete experimental experience. The laboratory experiments have an introduction for the students to read a pre-lab for them to complete before coming to the lab a data sheet to complete during the lab and a post-lab which would give them an opportunity to reinforce their understanding of the experiment completed Instructor resources include a list of all equipment and supplies needed for 24 students a lab preparation guide an answer key to all pre-lab and post-lab questions sample data for remote learners and a suggested rubric for grading the labs Additional features include Tested laboratory exercises with instructor resources for environmental science students Environmental calculations industrial regulation and environmental stewardship Classroom and remote exercises An excellent user friendly and thought provoking presentation which will appeal to students with little or no science background A qualitative approach to the chemistry behind many of our environmental issues today

Journal of Research of the National Bureau of Standards United States. National Bureau of Standards,1975

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Available, 1985 Study Package for Indian Air Force Airmen Group X & Y (Technical & Non Technical Trades)

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Semiannual with semiannual and annual indexes References to all scientific and technical literature coming from DOE its laboratories energy centers and contractors Includes all works deriving from DOE other related government sponsored information and foreign nonnuclear information Arranged under 39 categories e g Biomedical sciences basic studies Biomedical sciences applied studies Health and safety and Fusion energy Entry gives bibliographical information and abstract Corporate author subject report number indexes

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "Answers Key To Chemistry Radioactivity Packet 15," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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