

Unit Test Review

1. A sample of $\text{CO}_2(\text{s})$ and a sample of $\text{CO}_2(\text{g})$ differ in their

- A) chemical compositions
- B) empirical formulas
- C) molecular structures
- D) physical properties

2. At STP, which physical property of aluminum always remains the same from sample to sample?

- A) mass
- B) density
- C) length
- D) volume

3. Which two particle diagrams represent mixtures of diatomic elements?



- A) A and B
- B) A and C
- C) B and C
- D) B and D

4. Which substance can be broken down by chemical means?

- A) CO
- B) Ce
- C) Ca
- D) Cu

5. A 10.0-gram sample of $\text{H}_2\text{O}(\text{l})$ at 23.0°C absorbs 209 joules of heat. What is the final temperature of the $\text{H}_2\text{O}(\text{l})$ sample?

- A) 5.0°C
- B) 18.0°C
- C) 28.0°C
- D) 50.0°C

6. Object A, at 40°C and object B, at 80°C are placed in contact with each other. Which statement describes the heat flow between the objects?

- A) Heat flows from object A to object B.
- B) Heat flows from object B to object A.
- C) Heat flows in both directions between the objects.
- D) No heat flow occurs between the objects.

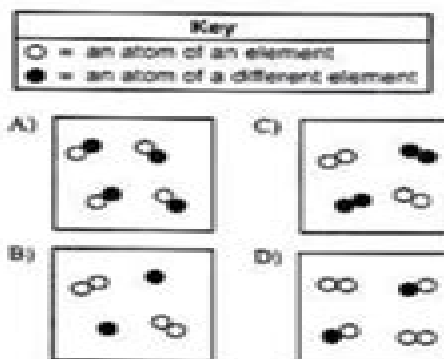
7. Which phase change results in the release of energy?

- A) $\text{H}_2\text{O}(\text{s}) \rightarrow \text{H}_2\text{O}(\text{l})$
- B) $\text{H}_2\text{O}(\text{s}) \rightarrow \text{H}_2\text{O}(\text{g})$
- C) $\text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_2\text{O}(\text{g})$
- D) $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$

8. The temperature of a sample of water changes from 10°C to 20°C when the water absorbs 100 calories of heat. What is the mass of the sample?

- A) 1 g
- B) 10 g
- C) 100 g
- D) 1000 g

9. Which particle diagram represents a mixture of an element and a compound?



10. Bronze contains 90 to 95 percent copper and 5 to 10 percent tin. Because these percentages can vary, bronze is classified as

- A) a compound
- B) an element
- C) a mixture
- D) a substance

11. Which process would most effectively separate two liquids with different molecular polarities?

- A) filtration
- B) fermentation
- C) distillation
- D) conductivity

Q#	answer #	Q#	answer #	Q#	answer #	Q#	answer #
1	D	4	A	7	D	10	C
2	B	5	C	8	B	11	C
3	B	6	B	9	D		

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