



THE PROPERTIES OF MATTER

Review Worksheet

1. What are the properties of matter? Properties of matter are the characteristics that distinguish one substance from another.

- a. What is mass? Mass is the amount of matter in an object. What is volume? Volume is the amount of space an object takes up. What is density? Density is the mass of an object divided by its volume.
- b. What are the three states of matter? Solid, liquid, and gas. What are the properties of solids? Solids have a fixed shape and volume. What are the properties of liquids? Liquids have a fixed volume but no fixed shape. What are the properties of gases? Gases have no fixed shape or volume.
- c. What are the three types of mixtures? Homogeneous, heterogeneous, and colloidal. What are the properties of homogeneous mixtures? Homogeneous mixtures have a uniform composition throughout. What are the properties of heterogeneous mixtures? Heterogeneous mixtures have a non-uniform composition. What are the properties of colloidal mixtures? Colloidal mixtures have particles that are too small to be seen with the naked eye but large enough to scatter light.
- d. What are the three types of pure substances? Element, compound, and mixture. What are the properties of elements? Elements are pure substances that cannot be broken down into simpler substances by chemical means. What are the properties of compounds? Compounds are pure substances that can be broken down into simpler substances by chemical means. What are the properties of mixtures? Mixtures are combinations of two or more pure substances that are not chemically combined.
- e. What are the three types of changes of state? Melting, freezing, and boiling. What are the properties of melting? Melting is the change from solid to liquid. What are the properties of freezing? Freezing is the change from liquid to solid. What are the properties of boiling? Boiling is the change from liquid to gas.
- f. What are the three types of mixtures? Homogeneous, heterogeneous, and colloidal. What are the properties of homogeneous mixtures? Homogeneous mixtures have a uniform composition throughout. What are the properties of heterogeneous mixtures? Heterogeneous mixtures have a non-uniform composition. What are the properties of colloidal mixtures? Colloidal mixtures have particles that are too small to be seen with the naked eye but large enough to scatter light.
- g. What are the three types of pure substances? Element, compound, and mixture. What are the properties of elements? Elements are pure substances that cannot be broken down into simpler substances by chemical means. What are the properties of compounds? Compounds are pure substances that can be broken down into simpler substances by chemical means. What are the properties of mixtures? Mixtures are combinations of two or more pure substances that are not chemically combined.

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