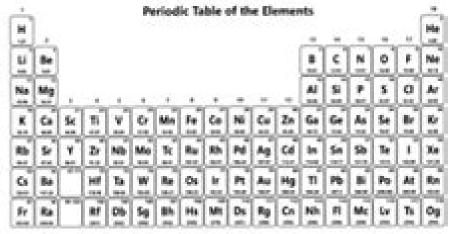
Name: Period:

Select the correct answers by circling one from each letter.

- 1. Which of the following in each pair has the smallest radius?
 - a. P O (circle one)
 - b. Te I (circle one)
 - c. Ne Kr (circle one)
 - d. Na S (circle one)
- 2. Which of the following in each group has the largest ionization energy?
 - a. S Cl Ar (circle one)
 - b, K Ca Rb (circle one)
 - c. Ar Xe Kr (circle one)
 - d. O N C (circle one)
- 3. Which has the highest electronegativity in each group?
 - a. B C N (circle one)
 - b. B Al Ge (circle one)
 - e. Ge O P (circle one)
 - d. Se S Cl (circle one)
- Name three metal <u>families</u> on the periodic table.
 - 3.
 - Ъ.





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M Carnoy

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