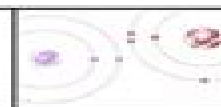


transform into a noble gas. As a result, there is a chemical connection because chlorine wants to take an electron while sodium wants to give one away.

Activity B:
Ionic compounds

Get the Gizmo ready:

- Click **Reset**. Turn off **Show charge**.
- Select **Lithium (Li)** and **Oxygen (O)**.



Question: How are ionic compounds formed?

1. **Observe:** Look at the purple lithium atom and the red oxygen atom. Recall that most atoms are stable when their outermost ring has eight electrons. Some atoms, such as lithium and beryllium, are stable when their outermost ring has two electrons.

A. How many electrons will the lithium atom give up to become stable?

1

B. How many electrons does the oxygen atom need to become stable?

2

C. Can a stable compound be made from these two atoms? Explain why or why not.

No, because lithium gave up one atom, oxygen will only have seven electrons when it requires eight, and lithium will only have seven if oxygen gave it seven electrons.

2. **Form bonds:** Click **Add metal** to add another lithium atom, and then transfer electrons from the lithium to the oxygen. Click **Check**.

A. Did you make a stable compound?

yes

B. Turn on **Show formula**. What is the formula of this compound?

Li_2O

C. Turn on **Show charge**. What is the charge of each ion? Li

1+

Li

1+

O

2-

3. **Practice:** Use the Gizmo to create stable compounds from the combinations given below. After transferring electrons, arrange the atoms to demonstrate the attraction between positively charged ions and negatively charged ions. Click **Check** to check each compound.

For each compound, click the **camera** (📷) icon to take a snapshot. Right-click the image, and click **Copy image**. Paste each image into a blank document to turn in with this worksheet. Write the ionic charges (such as Ca^{2+}) and chemical formulas below.

A. Lithium and fluorine:

Li 1+ F 1-

LiF

B. Beryllium and oxygen:

Be 2+ O 2-

BeO

C. Magnesium and fluorine:

Mg 1+ F 1-

MgF

Answer Key To Ionic Bonds Gizmo

Primrose Kitten

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Primrose Kitten' text.

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