

Teacher Answer Key Page 1 of 2

III. Changes Between States and Processes that **change** state

- A. **Freezing**- a substance changing from a **liquid** to **solid** state (the substance is **losing energy**)
The freezing point of water is 0°C - only water
- B. **Melting**- a substance changing from a **solid** to **liquid** state (the substance **is gaining energy**)
The melting point of water is 0°C -same as freezing point.
- C. **Boiling**- A substance changing from a liquid to a gas (the substance **is gaining more energy**)
The boiling point of water is 100°C .
- D. **Evaporation**- process when particles escape from a **liquid** and become a **gas**.
(the substance is **gaining energy**)
- E. **Condensation**- process when a **gas** becomes a **liquid**
Ex. hot steam on the cooler mirror after a hot shower (the substance is **losing energy**)

****A change in state of matter occurs when the substance gains or loses heat or energy.**

This is very important for the student to understand!!!!

Answer Key To Phases Of Matter Guided

George Graybill



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