

Name: KEY

15 Pts.

Chemistry Practice: Writing Chemical Formulas

Write a chemical formula for each substance.

- | | | | |
|--|--------------------------------|--|---------------------------|
| 1. <u>NaCl</u> | sodium chloride | 39. <u>Ca(NO₃)₂</u> | calcium nitrate |
| 2. <u>N₂O₅</u> | dinitrogen pentoxide | 40. <u>SO₃</u> | sulfur trioxide |
| 3. <u>H₂S</u> | hydrosulfuric acid | 41. <u>KCN</u> | potassium cyanide |
| 4. <u>K₂SO₄</u> | potassium sulfate | 42. <u>Pb(NO₃)₂</u> | lead(II) nitrate |
| 5. <u>H₂C₂O₄</u> | oxalic acid | 43. <u>H₂S</u> | hydrogen sulfide |
| 6. <u>Ag₂C₂H₃O₂</u> | silver acetate | 44. <u>CoCl₃</u> | cobalt(III) chloride |
| 7. <u>Cr(NO₃)₃</u> | chromium(III) nitrate | 45. <u>SF₆</u> | sulfur hexafluoride |
| 8. <u>HC₂H₃O₂</u> | acetic acid | 46. <u>Ca₃N₂</u> | calcium nitride |
| 9. <u>(NH₄)₂CO₃</u> | ammonium carbonate | 47. <u>CuI</u> | copper(I) iodide |
| 10. <u>Ca(OH)₂</u> | calcium hydroxide | 48. <u>SiO₂</u> | silicon dioxide |
| 11. <u>H₂C₄H₄O₆</u> | tartaric acid | 49. <u>Sn(C₂H₃O₂)₄</u> | tin(IV) acetate |
| 12. <u>Hg(NO₃)₂</u> | mercury(II) nitrate | 50. <u>CCl₄</u> | carbon tetrachloride |
| 13. <u>N₂O</u> | dinitrogen monoxide | 51. <u>CuS</u> | copper(II) sulfide |
| 14. <u>Fe₂O₃</u> | iron(III) oxide | 52. <u>Pb₃(PO₄)₂</u> | lead(II) phosphate |
| 15. <u>Pb(ClO₃)₂</u> | lead(II) chlorate | 53. <u>XeCl₄</u> | xenon tetrachloride |
| 16. <u>(NH₄)₃PO₄</u> | ammonium phosphate | 54. <u>Rb₂O</u> | rubidium oxide |
| 17. <u>ZnCl₂</u> | zinc chloride | 55. <u>MgSe</u> | magnesium selenide |
| 18. <u>Ca₃(PO₄)₂</u> | calcium phosphate | 56. <u>NH₄Cl</u> | ammonium chloride |
| 19. <u>OF₂</u> | oxygen difluoride | 57. <u>Fe(C₂H₃O₂)₃</u> | iron(III) acetate |
| 20. <u>NaN₃</u> | sodium azide | 58. <u>K₂Cr₂O₇</u> | potassium dichromate |
| 21. <u>Fe₂(SO₄)₃</u> | iron(III) sulfate | 59. <u>PBr₃</u> | phosphorous tribromide |
| 22. <u>H₃AsO₃</u> | arsenous acid | 60. <u>Na₃PO₃</u> | sodium phosphite |
| 23. <u>Cr₂O₃</u> | chromium(III) oxide | 61. <u>Na₃PO₄</u> | sodium phosphate |
| 24. <u>N₂O₄</u> | dinitrogen tetroxide | 62. <u>Hg(NO₃)₂</u> | mercury(II) nitrate |
| 25. <u>NH₄NO₃</u> | ammonium nitrate | 63. <u>LiHCO₃</u> | lithium bicarbonate |
| 26. <u>AsBr₃</u> | gold(III) bromide | 64. <u>CrF₃</u> | chromium(III) fluoride |
| 27. <u>CO</u> | carbon monoxide | 65. <u>PbI₂</u> | lead(II) iodide |
| 28. <u>K₂CO₃</u> | potassium carbonate | 66. <u>H₂SO₃</u> | sulfurous acid |
| 29. <u>HIO₃</u> | iodic acid | 67. <u>SnF₂</u> | tin(II) fluoride |
| 30. <u>CsCl</u> | cesium chloride | 68. <u>HgCrO₄</u> | mercury(II) chromate |
| 31. <u>Ni(MnO₄)₂</u> | nickel(II) permanganate | 69. <u>KNO₃</u> | potassium nitrate |
| 32. <u>Al₂(SO₄)₃</u> | aluminum sulfate | 70. <u>SrCl₂</u> | strontium chloride |
| 33. <u>Al₂(SO₃)₃</u> | aluminum sulfite | 71. <u>P₄O₁₀</u> | tetraphosphorous decoxide |
| 34. <u>Ba(C₂H₃O₂)₂</u> | barium acetate | 72. <u>KNO₃</u> | potassium nitrate |
| 35. <u>Mn(OH)₃</u> | manganese(III) hydroxide | 73. <u>KNO₂</u> | potassium nitrite |
| 36. <u>KH₂PO₄</u> | potassium dihydrogen phosphate | 74. <u>K₃N</u> | potassium nitride |
| 37. <u>HF</u> | hydrofluoric acid | 75. <u>CaO</u> | calcium oxide |
| 38. <u>Al(BrO₃)₃</u> | aluminum bromate | 76. <u>Fe(IO₄)₂</u> | iron(II) periodate |

Answer Key For Chemistry Lab Names And Formulas

Jo Allan Beran



Answer Key For Chemistry Lab Names And Formulas:

Reviewing **Answer Key For Chemistry Lab Names And Formulas**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Answer Key For Chemistry Lab Names And Formulas**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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