

Periodic Trends Worksheet

Directions: Use your notes to answer the following questions.

1. Rank the following elements by increasing atomic radius: carbon, aluminum, oxygen, potassium.

Oxygen < Carbon < Aluminum < Potassium

2. Rank the following elements by increasing electronegativity: sulfur, oxygen, neon, aluminum.

Neon < Aluminum < Sulfur < Oxygen

3. Why does fluorine have a higher ionization energy than iodine?

Fluorine has nine protons attracting 9 electrons which are much closer to the nucleus compared to iodine. This results in smaller size of Fluorine than iodine making it difficult to remove an electron from the atom.

4. Why do elements in the same family generally have similar properties?

Because they have same number of electrons in the outer shell (valence electrons) which take part in chemical reaction.

5. Indicate whether the following properties increase or decrease from left to right across the periodic table.

- | | |
|--|-----------|
| a. atomic radius (excluding noble gases) | Decreases |
| b. first ionization energy | Increases |
| c. electronegativity | Increases |

6. What trend in atomic radius occurs down a group on the periodic table? What causes this trend?

Atomic radius **increases** down the group on the periodic table. As we compare the elements down the group, the effective nuclear charge increases, but at the same time the outermost electrons are found in the shell that is farther away from the nucleus. Also the number of screening electrons increases. This results in reduced attraction between the nucleus and the outermost electrons.

7. What trend in ionization energy occurs across a period on the periodic table? What causes this trend?

Ionization energy **increases** from left to right across the period. The atomic size becomes smaller from left to right. So it becomes harder to remove electron from the atom. Hence the energy required to do so (ionization Energy) increases.

Answer Key For Periodic Table Trend Packet

VM Jensen



Answer Key For Periodic Table Trend Packet:

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Table of Contents Answer Key For Periodic Table Trend Packet

1. Understanding the eBook Answer Key For Periodic Table Trend Packet
 - The Rise of Digital Reading Answer Key For Periodic Table Trend Packet
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Periodic Table Trend Packet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Periodic Table Trend Packet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Periodic Table Trend Packet
 - Personalized Recommendations
 - Answer Key For Periodic Table Trend Packet User Reviews and Ratings
 - Answer Key For Periodic Table Trend Packet and Bestseller Lists

5. Accessing Answer Key For Periodic Table Trend Packet Free and Paid eBooks
 - Answer Key For Periodic Table Trend Packet Public Domain eBooks
 - Answer Key For Periodic Table Trend Packet eBook Subscription Services
 - Answer Key For Periodic Table Trend Packet Budget-Friendly Options
6. Navigating Answer Key For Periodic Table Trend Packet eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key For Periodic Table Trend Packet Compatibility with Devices
 - Answer Key For Periodic Table Trend Packet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Periodic Table Trend Packet
 - Highlighting and Note-Taking Answer Key For Periodic Table Trend Packet
 - Interactive Elements Answer Key For Periodic Table Trend Packet
8. Staying Engaged with Answer Key For Periodic Table Trend Packet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Periodic Table Trend Packet
9. Balancing eBooks and Physical Books Answer Key For Periodic Table Trend Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Periodic Table Trend Packet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key For Periodic Table Trend Packet
 - Setting Reading Goals Answer Key For Periodic Table Trend Packet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key For Periodic Table Trend Packet
 - Fact-Checking eBook Content of Answer Key For Periodic Table Trend Packet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web when addressing something like osmosis it is really another form of diffusion for water but flipped in diffusion we don't see the polarity size of molecules or charge playing a role

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web the question often arises as to what is the difference between osmosis and diffusion which are two forms of biological transport osmosis is the movement of solvent

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web osmosis is a chemical process of absorption or diffusion of a solvent through a semi permeable membrane as of a living cell to a higher concentration of solute and then it

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web osmosis is the diffusion of a solvent through a differentially permeable membrane in biological systems the solvent will usually be water osmosis will occur whenever the

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web transport in cells for an organism to function substances must move into and out of cells three processes contribute to this movement diffusion osmosis and active transport

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web in cells some molecules can move down their concentration gradients by crossing the lipid portion of the membrane directly while others must pass through membrane proteins in

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web why can generally only very small hydrophobic molecules across the cell membrane by simple diffusion explain how facilitated diffusion assists in osmosis in cells be sure

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web aug 3 2023 in this enlightening article we will explore the significance of the ap biology laboratory 1 diffusion and osmosis answer key its alignment with essential

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web label the diagrams of cells using the following terms diffusion active transport osmosis facilitated diffusion or equilibrium the arrows show the direction of transport you may

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web 1 how are the molecules moving in the examples below write osmosis or diffusion a the student sitting next to you just came from gym class and forgot to shower and you

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web movement of molecules from high to low concentrate diffusion a molecule composed of two hydrogen and one oxygen water a solution that has equal amounts of particles

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web 1 define diffusion 2 what is moving during osmosis 3 which type of cellular transport requires energy passive transport or active transport 4 what are two types of

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web there are two major ways that molecules can be moved across a membrane and the distinction has to do with whether or not cell energy is used passive mechanisms like diffusion use no energy while active transport requires energy to get done