TEACHING TRANSPARENCY WORKSHEET

Types of Matter

Use with Chapter 3, Section 3.4

- 1. Into what two broad classes can all matter be divided?
- 2. What is the difference between a mixture and a pure substance?
- 3. What is the difference between a compound and an element?
- Can a compound be a heterogeneous mixture? Explain your answer by referring to the diagram.
- 5. A list of compounds and elements is given below. Circle the substances that are elements.

gold:

aluminum

weater

sugar

oxygen

platinum brass

- salt chlorine
- 6. How can you tell the difference between a homogeneous mixture and a heterogeneous mixture?
- 7. Label each mixture below as either homogeneous or heterogeneous.
 - a. air

e. finger paint ____

b. clay ____

- f. vinegar _____
- c. homemade lemonade (with pulp)
- q. soil ____

- d. oatmeal raisin cookie
- List three methods that are commonly used to separate mixtures into their component substances.

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Chemistry: Matter and Change • Chapter 3

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