

Naming Ionic Compounds

What are the structural units that make up ionic compounds and how are they named?

Why?

When working in chemistry, it is often convenient to write a chemical in symbols. For example we might write down the substance table salt as NaCl . In talking about chemistry however, it is a bit tacky to say "en-ay see-ell" when we want to refer to a substance. Also, in formal writing we should use the name of the compound rather than its symbols. Therefore we need to learn how to say the proper names of ionic substances.

Model 1 – Ion Charges for Selected Elements

1	H ⁺												
2	Li ⁺	Be ²⁺							N ³⁻	O ²⁻	F ⁻		
3	Na ⁺	Mg ²⁺	Transition elements				Al ³⁺		P ³⁻	S ²⁻	Cl ⁻		
4	K ⁺	Ca ²⁺	Fe ²⁺ Fe ³⁺	Ni ²⁺ Ni ³⁺	Cu ⁺ Cu ²⁺	Zn ²⁺						Br ⁻	
5	Rb ⁺	Sr ²⁺			Ag ¹⁺			Sn ²⁺ Sn ⁴⁺				I ⁻	
6		Ba ²⁺				Hg ₂ ²⁺ Hg ²⁺		Pb ²⁺ Pb ⁴⁺					
	Cations							Anions					

1. Based on the information in Model 1:

- a. Identify three elements that form only one cation. $\leftarrow$ Positive Ion
 Na^+ , Sr^{2+} , Al^{3+} (Any box with one element works.)
- b. Identify three elements that form only one anion. $\leftarrow$ Negative Ion
 P^{3-} , O^{2-} , Br^- (Any box with one element works.)
- c. Identify three elements that form more than one cation.
 Fe^{+2} and Fe^{+3} , Ni^{+2} and Ni^{+3} , Cu^{+1} and Cu^{+2}
- d. In what region of the periodic table are these "multiple ion" elements usually located?
Transition elements or "d" block

2. Consider the ions of potassium (K) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of potassium (K) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.
- $2K^+$ and S^{2-} or K_2S
3. Consider the ions of iron (Fe) and sulfur (S). Write chemical formulas for all possible ionic compounds involving these ions, using the simplest ratio(s) of iron (Fe) and sulfur (S). Keep in mind that the sum of the charges in an ionic compound must equal zero.



Answer Key For Naming Ionic Compounds Pogil

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1. Understanding the eBook Answer Key For Naming Ionic Compounds Pogil
 - The Rise of Digital Reading Answer Key For Naming Ionic Compounds Pogil
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Naming Ionic Compounds Pogil
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Naming Ionic Compounds Pogil
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Naming Ionic Compounds Pogil
 - Personalized Recommendations
 - Answer Key For Naming Ionic Compounds Pogil User Reviews and Ratings
 - Answer Key For Naming Ionic Compounds Pogil and Bestseller Lists
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- Answer Key For Naming Ionic Compounds Pogil eBook Subscription Services
- Answer Key For Naming Ionic Compounds Pogil Budget-Friendly Options
- 6. Navigating Answer Key For Naming Ionic Compounds Pogil eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key For Naming Ionic Compounds Pogil Compatibility with Devices
 - Answer Key For Naming Ionic Compounds Pogil Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Naming Ionic Compounds Pogil
 - Highlighting and Note-Taking Answer Key For Naming Ionic Compounds Pogil
 - Interactive Elements Answer Key For Naming Ionic Compounds Pogil
- 8. Staying Engaged with Answer Key For Naming Ionic Compounds Pogil
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Naming Ionic Compounds Pogil
- 9. Balancing eBooks and Physical Books Answer Key For Naming Ionic Compounds Pogil
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Naming Ionic Compounds Pogil
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key For Naming Ionic Compounds Pogil
 - Setting Reading Goals Answer Key For Naming Ionic Compounds Pogil
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key For Naming Ionic Compounds Pogil
 - Fact-Checking eBook Content of Answer Key For Naming Ionic Compounds Pogil
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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