

# Osmosis and Tonicity

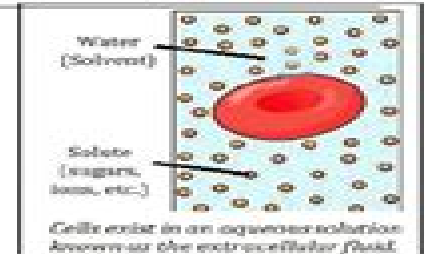
All cells are surrounded by a solution called the **extracellular fluid**. Cells obtain the nutrients/water they need from this fluid, and also expel their waste into it.

The extracellular fluid is an **aqueous solution**

- Water is the solvent
- Sugars (glucose), gases ( $O_2$ ,  $CO_2$ ), and ions ( $Na^+$ ,  $Cl^-$ ) are common solutes

The movement of many of these solutes is actively controlled/regulated by the cell.

The movement of the solvent, water, is **NOT** regulated. Water moves freely across a cell membrane via **osmosis**.



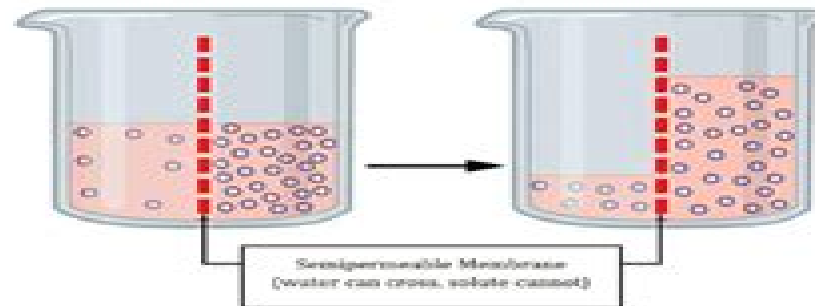
*This video explains osmosis very clearly. I highly recommend watching it.*

<http://tinyurl.com/q8aggr2>

## Osmosis

Osmosis is the diffusion of water across a semipermeable membrane **from an area of low solute concentration to an area of high solute concentration**

- The water always moves towards the area of higher solute concentration, until the concentrations on both sides of the membrane are equal.
- Once the concentration is balanced on both sides, there is no net movement of water. The water is crossing the membrane in both directions at an equal rate. This is known as a **dynamic equilibrium**.

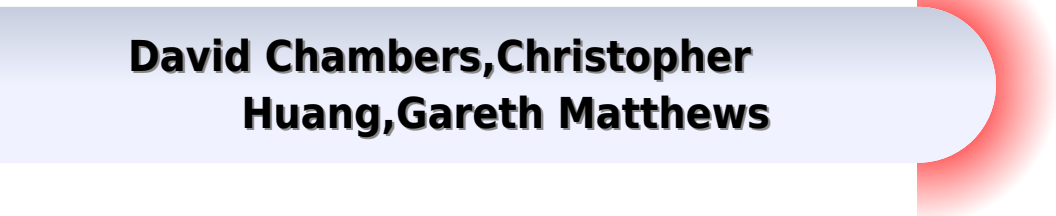


**Tonicity** compares the relative concentration of the intracellular fluid (solution inside the cell) and extracellular fluid (solution outside of the cell).

- Comparing the inside/outside concentrations allows us to predict whether water will move into the cell, out of the cell, or remain in a dynamic equilibrium.

# Answer Key For Osmosis And Tonicity

**David Chambers, Christopher  
Huang, Gareth Matthews**



## **Answer Key For Osmosis And Tonicity:**

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## **Table of Contents Answer Key For Osmosis And Tonicity**

1. Understanding the eBook Answer Key For Osmosis And Tonicity
  - The Rise of Digital Reading Answer Key For Osmosis And Tonicity
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Osmosis And Tonicity
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key For Osmosis And Tonicity
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Osmosis And Tonicity
  - Personalized Recommendations
  - Answer Key For Osmosis And Tonicity User Reviews and Ratings
  - Answer Key For Osmosis And Tonicity and Bestseller Lists
5. Accessing Answer Key For Osmosis And Tonicity Free and Paid eBooks
  - Answer Key For Osmosis And Tonicity Public Domain eBooks
  - Answer Key For Osmosis And Tonicity eBook Subscription Services
  - Answer Key For Osmosis And Tonicity Budget-Friendly Options
6. Navigating Answer Key For Osmosis And Tonicity eBook Formats
  - ePub, PDF, MOBI, and More
  - Answer Key For Osmosis And Tonicity Compatibility with Devices
  - Answer Key For Osmosis And Tonicity Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key For Osmosis And Tonicity
  - Highlighting and Note-Taking Answer Key For Osmosis And Tonicity
  - Interactive Elements Answer Key For Osmosis And Tonicity
8. Staying Engaged with Answer Key For Osmosis And Tonicity

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answer Key For Osmosis And Tonicity
- 9. Balancing eBooks and Physical Books Answer Key For Osmosis And Tonicity
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key For Osmosis And Tonicity
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key For Osmosis And Tonicity
  - Setting Reading Goals Answer Key For Osmosis And Tonicity
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key For Osmosis And Tonicity
  - Fact-Checking eBook Content of Answer Key For Osmosis And Tonicity
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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