

Sublevels

- Each orbital type holds a different number of electrons
 - S shape has 1 orbital for a total of 2 electrons
 - P shape has 3 orbitals for a total of 6 electrons
 - D shape has 5 orbitals for a total of 10 electrons
 - F shape has 7 orbitals for a total of 14 electrons

14 Spag Sublevels

CL Gary



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