1. Which best describes the current atomic theory?

 a. atoms consist of electrons circling in a definite orbits around a positive nucleus.

b. atoms are composed of electrons in a cloud aound a positive nucleus.

c. Atoms can easily be split, at which time they become radioactive.

d. an atom's mass is determined by the mass of its neutrons.

- 2. What is the nuclear composition of uranium-235?
 - a. 92 electrons + 143 protons e. 143 protons + 92 neutrons
 - b. 92 protons + 143 electrons d. 92 protons + 143 neutrons

3. Which best describes the relationship between subutomic particles in any neutral atom?

- a, the number of protons equals the number of electrons
- b, the number of protons equals the number of neutrons
- c. the number of neutrons equals the numbers of electrons
- d. the number of neutrons is greater than the number of protons
- 4. What is the name of the compound PbO₂?
 - a. lead oxide b. lead(II) oxide c. lead oxide (I) d. lead (IV) oxide
- 5. What is the name of HCl(aq)?
 - a, chloric acid b, bydrochloric acid c, bydrogen chloride d, perchloric acid
- 6. What is the formula for calcium nitrate?
 - a. $CaNO_3$ b. $Ca(NO_2)_2$ c. $Ca(NO_3)_2$ d. Ca_3N_2
- 7. What is the correct formula for dinitrogen pentoxide?
 - a. NaO b. NO₁ e. N₂O₂ d. NO₄

8. If the volume of 18.5 gram piece of metal is $2.35~\mathrm{cm}^3$, what is the identity of the metal?

a. iron b. lead c. nickel d. zinc

Which substance listed in the table is a liquid at 27°C?		Melting Point	Beiling Point
a. I b. II c. III d. IV	1	28°C	140°C
	111	-10°C	28°C
	1111	20°C	140°C
	IV	790°C	14 %

- 10. Which will increase the solubility of MOST solid solutes?
- a, decreasing the temperature
- b, decreasing the amount of solvent at constant temperature
- c. increasing the amount of solute at constant temperature
- d. increasing the temperature

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