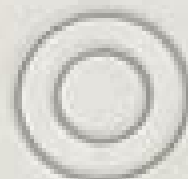


15. Calculate the missing information and then draw the Bohr Diagram and Lewis Structure for each element.



Li

Atomic # = 3

Mass # = 7

of P = ____

of N = ____

of E = ____



Ne

Atomic # = 10

Mass # = 20

of P = ____

of N = ____

of E = ____



Mg

Atomic # = 12

Mass # = 24

of P = ____

of N = ____

of E = ____



Cl

Atomic # = 17

Mass # = 35

of P = ____

of N = ____

of E = ____



He

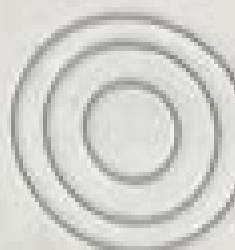
Atomic # = 2

Mass # = 4

of P = ____

of N = ____

of E = ____



Si

Atomic # = 14

Mass # = 28

of P = ____

of N = ____

of E = ____

16. Answer the questions below based on the elements in question #15.

(1) Which elements had a filled outermost shell? ____

(2) Which element would be most likely to lose electrons in a chemical bond? ____

(3) Which element would be most likely to gain electrons in a chemical bond? ____

(4) Which elements are not likely to bond with other elements? ____ Why? ____

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