

Analytical Chemistry MCQs: TYBSC SEM III

8. The solubility product K_{sp} of sparingly soluble Ag_2CrO_4 is 4×10^{-12} . The molar solubility of the salt is
- (a) 1.0×10^{-4} moles/lit
 - (b) 2.0×10^{-4} moles/lit
 - (c) 1.0×10^{-5} moles/lit
 - (d) 1.2×10^{-12} moles/lit
9. Solubility product of $AgCl$ is 1.44×10^{-4} moles/lit at 373K. The solubility of the $AgCl$ in boiling of water will be....
- (a) 0.72×10^{-4} moles/lit
 - (b) 0.72×10^{-2} moles/lit
 - (c) 1.20×10^{-2} moles/lit
 - (d) 1.20×10^{-4} moles/lit
10. Precipitation takes place when the ionic product....
- (a) Is zero
 - (b) Exceeds the solubility product
 - (c) Equal to solubility product
 - (d) Is less than the solubility product
11. The electromotive force is expressed in the unit of
- (a) Ohm
 - (b) Volt
 - (c) Ampere
 - (d) Coulomb
12. If the cell can provide electrical energy to an external system, it is called as
- (a) Galvanic cell
 - (b) Electrolytic cell
 - (c) Concentration cell
 - (d) Chemical cell
13. The current per unit area of electrode surface is known as....
- (a) Current efficiency
 - (b) Current yield
 - (c) Current density
 - (d) None of these

Analytical Chemistry Mcqs

Amar Bhutani



Analytical Chemistry Mcqs:

MCQs Chemistry V. K. Ahluwalia, 2013-12-30 This book has been divided in 37 chapters under three parts Inorganic Chemistry Organic Chemistry and Physical Chemistry for convenient understanding It also includes solved model test papers of the previous three years of AIIMS CBSE PMT CPMT UP to enable students to develop the skills of problem solving and time management essential for any entrance examination In addition to providing answers to all the questions detailed explanatory notes to selected difficult questions have also been provided to justify the answer A separate section of Assertions and Reasons is also given at the end of each chapter Exhaustive Question Bank Explanatory Notes and Hints Assertions Reasons Includes Pre solved papers of five years Models Test Papers of AIIMS CBSE PMT CPMT **Class 11-12**

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Table of Contents Analytical Chemistry Mcqs

1. Understanding the eBook Analytical Chemistry Mcqs
 - The Rise of Digital Reading Analytical Chemistry Mcqs
 - Advantages of eBooks Over Traditional Books
2. Identifying Analytical Chemistry Mcqs
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Analytical Chemistry Mcqs
 - User-Friendly Interface
4. Exploring eBook Recommendations from Analytical Chemistry Mcqs
 - Personalized Recommendations
 - Analytical Chemistry Mcqs User Reviews and Ratings

- Analytical Chemistry Mcqs and Bestseller Lists
- 5. Accessing Analytical Chemistry Mcqs Free and Paid eBooks
 - Analytical Chemistry Mcqs Public Domain eBooks
 - Analytical Chemistry Mcqs eBook Subscription Services
 - Analytical Chemistry Mcqs Budget-Friendly Options
- 6. Navigating Analytical Chemistry Mcqs eBook Formats
 - ePub, PDF, MOBI, and More
 - Analytical Chemistry Mcqs Compatibility with Devices
 - Analytical Chemistry Mcqs Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Analytical Chemistry Mcqs
 - Highlighting and Note-Taking Analytical Chemistry Mcqs
 - Interactive Elements Analytical Chemistry Mcqs
- 8. Staying Engaged with Analytical Chemistry Mcqs
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Analytical Chemistry Mcqs
- 9. Balancing eBooks and Physical Books Analytical Chemistry Mcqs
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Analytical Chemistry Mcqs
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Analytical Chemistry Mcqs
 - Setting Reading Goals Analytical Chemistry Mcqs
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Analytical Chemistry Mcqs
 - Fact-Checking eBook Content of Analytical Chemistry Mcqs
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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