

Chapter 23, Section 1: The French Revolution Begins

Use your *Modern World History: Patterns of Interaction* textbook to read chapter 23.1 and answer the sections below. **DO NOT** copy your answers directly from the textbook. Instead, put your answers into your own words and in complete sentences so you can be sure to comprehend the historical material.



Terms and Names

For each term or name, write a brief description explaining their significance to the chapter's reading.

<i>Term or Name</i>	<i>Brief Description</i>
<i>Old Regime</i>	
<i>estate</i>	
<i>Estates-General</i>	
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Answer Key Guided The French Revolution Begins

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dissoiutl n 30d g of water w 4 which salt shows the least change

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web solvents the larger volume of the mixture and solutes the smaller volume of the mixture for example a hot chocolate is a solution in which the solute the chocolate powder is dissolved in the solvent the milk or water the solute and solvent can be

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web they should be able to come up with the following answers stir the water increase the temperature and increase the amount of water alternately you may challenge the students to a contest

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web use a solubility curve to determine whether a solution is saturated or unsaturated and to estimate the amount of excess solute that remains undissolved in the solution

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web give results of all calculations in blue cells you must put your answers in the indicated cells as the grade book pulls your data from those cells do not put any units in the number cells use the unit boxes given graph tab transfer from data page the temp and concentration of salt in units of g salt 100 g water

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solubility of KClO_3 unsaturated saturation and solubility curves in this experiment we expressed concentration in units of g
solute g H_2O putting concentrations in g solute g solution would be tedious and impractical see fig 2

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at 40 °C 200 cm³ of saturated lead (II) nitrate solution was prepared at a temperature of 90 °C

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analytical balance to measure mass set up an experimental work station to measure the solubility of a salt in water as a
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resembles the sloping solubility curve of potassium nitrate temperature solubility g 100 ml 10 °C 36 g 100 ml 20 °C 36 g 100 ml
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temperature in this lab 5 ml of water was used assume 1 0 g ml of water take the collected data make it proportional to get
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