

Identifying the 5 Types of Chemical ReactionsIdentify each of the reactions below as one of the 5 types of Reactions (Rxn):

- Synthesis Reaction
- Decomposition Reaction
- Single Replacement Reaction
- Double Replacement Reaction
- Combustion Reaction

Reactions (identify pattern below)Name of Reaction Type

|     |                                                                                                             |               |
|-----|-------------------------------------------------------------------------------------------------------------|---------------|
| 1.  | $2 \text{Na} + \text{Cl}_2 \rightarrow 2 \text{NaCl}$<br>EX:     A       B       →   AB     (A & B combine) | Ex: synthesis |
| 2.  | $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$                                           |               |
| 3.  | $\text{P}_4 + 3 \text{O}_2 \rightarrow 2 \text{P}_2\text{O}_3$                                              |               |
| 4.  | $2 \text{NO}_2 \rightarrow 2 \text{O}_2 + \text{N}_2$                                                       |               |
| 5.  | $\text{Na}_3\text{PO}_4 + 3 \text{KOH} \rightarrow 3 \text{NaOH} + \text{K}_3\text{PO}_4$                   |               |
| 6.  | $\text{C}_3\text{H}_6\text{O} + 4 \text{O}_2 \rightarrow 3 \text{CO}_2 + 3 \text{H}_2\text{O}$              |               |
| 7.  | $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2 \text{LiCl}$                          |               |
| 8.  | $\text{C}_6\text{H}_{12} + 9 \text{O}_2 \rightarrow 6 \text{CO}_2 + 6 \text{H}_2\text{O}$                   |               |
| 9.  | $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$                                                        |               |
| 10. | $2 \text{AgNO}_3 + \text{Cu} \rightarrow \text{Cu}(\text{NO}_3)_2 + 2 \text{Ag}$                            |               |
| 11. | $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$                                                |               |



# Answer Key To Chemical Reactions

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