

Chapter 5 The Periodic Table

Section 5.1 Organizing the Elements

(pages 126–129)

This section explains how *Mendeleev* organized elements into a periodic table. It also discusses the predictions he made about undiscovered elements and later the discovery of those elements supported his version of the table of the table.

Reading Strategy (page 126)

Identifying Main Ideas As you read, complete the table by identifying the main idea for each topic. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Topic	Main Idea
Mendeleev's proposal	Mendeleev arranged the elements in order of increasing mass so that elements with similar properties were in the same column.
Mendeleev's prediction	Mendeleev used the properties of existing elements to predict properties of undiscovered elements.
Evidence supporting Mendeleev's table	The close match between Mendeleev's predictions and the actual properties of new elements showed that his periodic table could be.

The Search for Order (page 126)

1. Is the following sentence true or false? The first elements to be identified were mainly gases. false
2. As the number of known elements grew, so did the need to organize them into groups based on their properties.
3. Circle the letter of each category that the French chemist Antoine Lavoisier used to classify elements.

☒ a. gases
c. liquids

☒ b. metals
☒ d. nonmetals

Mendeleev's Periodic Table (pages 127–128)

4. Is the following sentence true or false? Mendeleev needed to organize information about 63 elements. true
5. Mendeleev's strategy for classifying elements was modeled on a(n) card game or solitaire.
6. Circle the letter of each type of information Mendeleev knew about each element.

☒ a. name

☐ b. number of protons

☒ c. relative mass

☒ d. properties

Answer Key To Organizing Elements

Liying Dong



Answer Key To Organizing Elements:

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Table of Contents Answer Key To Organizing Elements

1. Understanding the eBook Answer Key To Organizing Elements
 - The Rise of Digital Reading Answer Key To Organizing Elements
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Organizing Elements
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Organizing Elements
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Organizing Elements
 - Personalized Recommendations
 - Answer Key To Organizing Elements User Reviews and Ratings
 - Answer Key To Organizing Elements and Bestseller Lists
5. Accessing Answer Key To Organizing Elements Free and Paid eBooks
 - Answer Key To Organizing Elements Public Domain eBooks
 - Answer Key To Organizing Elements eBook Subscription Services
 - Answer Key To Organizing Elements Budget-Friendly Options
6. Navigating Answer Key To Organizing Elements eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key To Organizing Elements Compatibility with Devices
 - Answer Key To Organizing Elements Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key To Organizing Elements
 - Highlighting and Note-Taking Answer Key To Organizing Elements
 - Interactive Elements Answer Key To Organizing Elements
8. Staying Engaged with Answer Key To Organizing Elements

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Answer Key To Organizing Elements
- 9. Balancing eBooks and Physical Books Answer Key To Organizing Elements
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Organizing Elements
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key To Organizing Elements
 - Setting Reading Goals Answer Key To Organizing Elements
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key To Organizing Elements
 - Fact-Checking eBook Content of Answer Key To Organizing Elements
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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