

Chemistry - Review Chapters 4-5: 6

Directions: Answer each of the following questions. You need not use complete sentences.

1. By what property did Mendeleev arrange the elements?

ATOMIC MASS

2. By what property did Moseley suggest that the periodic table be arranged?

ATOMIC NUMBER

3. What is a period? How many are there in the periodic table?

A HORIZONTAL ROW IN THE PERIODIC TABLE; 7

4. What is a group (also called a family)? How many are there in the periodic table?

A VERTICAL COLUMN IN THE PERIODIC TABLE; 18

5. State the number of valence electrons in an atom of:

a. sulfur	b. calcium	c. chlorine	d. arsenic
6	2	7	5

6. Give the names and chemical symbols for the elements that correspond to these atomic numbers:

a. 10	b. 18	c. 36	d. 90
Ne, NEON	Ar, ARGON	Kr, KRYPTON	Th, THORIUM

7. List, by number, both the period and group of each of these elements.

	<u>Symbol</u>	<u>Period</u>	<u>Group</u>
a. beryllium	Be	2	2
b. iron	Fe	4	8
c. lead	Pb	6	14

8. Which of the following pairs of elements belong to the same period?

a. Na and Cl	b. Na and Li	c. Na and Cu	d. Na and Ne
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9. Which of the following pairs of elements belong to the same group?

a. H and He	b. Li and Be	c. C and Pb	d. Ga and Ge
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10. How does an element's period number relate to the number of the energy level of its valence electrons?

PERIOD NUMBER = ENERGY LEVEL OF VALENCE ELECTRONS

11. What are the transition elements?

GROUPS 3-12

12. In what type of orbitals are the actinide and lanthanide electrons found?

f ORBITALS

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1. Understanding the eBook Answer Key To Prentice Hall Chemistry Guided
 - The Rise of Digital Reading Answer Key To Prentice Hall Chemistry Guided
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Prentice Hall Chemistry Guided
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Prentice Hall Chemistry Guided
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Prentice Hall Chemistry Guided

- Personalized Recommendations
 - Answer Key To Prentice Hall Chemistry Guided User Reviews and Ratings
 - Answer Key To Prentice Hall Chemistry Guided and Bestseller Lists
5. Accessing Answer Key To Prentice Hall Chemistry Guided Free and Paid eBooks
- Answer Key To Prentice Hall Chemistry Guided Public Domain eBooks
 - Answer Key To Prentice Hall Chemistry Guided eBook Subscription Services
 - Answer Key To Prentice Hall Chemistry Guided Budget-Friendly Options
6. Navigating Answer Key To Prentice Hall Chemistry Guided eBook Formats
- ePub, PDF, MOBI, and More
 - Answer Key To Prentice Hall Chemistry Guided Compatibility with Devices
 - Answer Key To Prentice Hall Chemistry Guided Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Answer Key To Prentice Hall Chemistry Guided
 - Highlighting and Note-Taking Answer Key To Prentice Hall Chemistry Guided
 - Interactive Elements Answer Key To Prentice Hall Chemistry Guided
8. Staying Engaged with Answer Key To Prentice Hall Chemistry Guided
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Prentice Hall Chemistry Guided
9. Balancing eBooks and Physical Books Answer Key To Prentice Hall Chemistry Guided
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Prentice Hall Chemistry Guided
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key To Prentice Hall Chemistry Guided
- Setting Reading Goals Answer Key To Prentice Hall Chemistry Guided
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key To Prentice Hall Chemistry Guided

- Fact-Checking eBook Content of Answer Key To Prentice Hall Chemistry Guided
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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