

Unit 1 Study Guide — ANSWER KEY

1. Define scientific method: **a procedure for collecting, testing, and communicating information**
2. Define hypothesis: **a suggested answer to the question, an educated guess**
3. Define experiment: **a scientific test so see what will happen**
4. Define dependent variable: **the variable that you do NOT control. (on a graph is usually on the Y axis)**
5. Define independent variable: **the variable that I control. (on a graph is usually on the X axis)**
6. Define control: **the part of the experiment where everything is left the same.**
7. Define observation: **information collected with any of your 5 senses**
8. Define density: **the ratio of mass per unit volume (D = m/v)**
9. Define accuracy: **the measure of how close your data is to the ACTUAL or true value.**
10. Define precision: **a measure of how close your data (measurements) are to each other.**
11. Why do scientists use SI units of measurement? **So that measurements can be understood internationally (around the world).**
12. List (in the correct order) the steps of the scientific method.

- | | |
|------------------------|-----------------------------------|
| 1. make an observation | 4. test hypothesis (experiment) |
| 2. ask a question | 5. analyze data/form a conclusion |
| 3. form a hypothesis | 6. develop a theory |

13. What mnemonic device is used to remember the metric system?

K H Da B d c m
g, m, L

14. What is the sentence (mnemonic) that will help you remember the METRIC prefixes in order?

King Henry Died By Drinking Chocolate Milk

15. Fill in the table below describing the metric system:

	Kilo	Hecto	Deca	Base	Deci	Centi	Milli
Abbreviation	K	H	Da	g, m, L	d	c	m
Meaning in powers of 10	10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}

Convert the following:

- | | |
|---------------------------------|------------------------------------|
| 16. 25L to <u>25000</u> mL | 21. 25 cm to <u>0.25</u> m |
| 17. 2.3 km to <u>230,000</u> cm | 22. 56784 mm to <u>0.056784</u> km |
| 18. 0.045 g to <u>45</u> mg | 23. 15 g to <u>0.015</u> kg |
| 19. 0.35 kg to <u>350</u> g | 24. 3.452 kL to <u>3452</u> L |
| 20. 300 mL to <u>0.300</u> L | 25. 47 m to <u>47000</u> mm |

Scientific Notation:

Convert the following to scientific notation:

- | | | | |
|------------|--------------------------------------|------------|--|
| 26) 0.005 | <u>5×10^{-3}</u> | 30) 0.025 | <u>2.5×10^{-2}</u> |
| 27) 5.050 | <u>5.05×10^1</u> | 31) 0.0025 | <u>2.5×10^{-3}</u> |
| 28) 0.0008 | <u>8×10^{-4}</u> | 32) 500 | <u>5×10^2</u> |
| 29) 1,000 | <u>1×10^3</u> | 33) 5,000 | <u>5×10^3</u> |

Answer Key Crct Study Guide 5

Huangqi Zhang



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Answer Key Crct Study Guide 5 Book Review: Unveiling the Power of Words

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Table of Contents Answer Key Crct Study Guide 5

1. Understanding the eBook Answer Key Crct Study Guide 5
 - The Rise of Digital Reading Answer Key Crct Study Guide 5
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key Crct Study Guide 5
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key Crct Study Guide 5
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key Crct Study Guide 5
 - Personalized Recommendations
 - Answer Key Crct Study Guide 5 User Reviews and Ratings
 - Answer Key Crct Study Guide 5 and Bestseller Lists
5. Accessing Answer Key Crct Study Guide 5 Free and Paid eBooks

- Answer Key Crct Study Guide 5 Public Domain eBooks
- Answer Key Crct Study Guide 5 eBook Subscription Services
- Answer Key Crct Study Guide 5 Budget-Friendly Options
- 6. Navigating Answer Key Crct Study Guide 5 eBook Formats
 - ePub, PDF, MOBI, and More
 - Answer Key Crct Study Guide 5 Compatibility with Devices
 - Answer Key Crct Study Guide 5 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key Crct Study Guide 5
 - Highlighting and Note-Taking Answer Key Crct Study Guide 5
 - Interactive Elements Answer Key Crct Study Guide 5
- 8. Staying Engaged with Answer Key Crct Study Guide 5
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key Crct Study Guide 5
- 9. Balancing eBooks and Physical Books Answer Key Crct Study Guide 5
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key Crct Study Guide 5
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key Crct Study Guide 5
 - Setting Reading Goals Answer Key Crct Study Guide 5
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key Crct Study Guide 5
 - Fact-Checking eBook Content of Answer Key Crct Study Guide 5
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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