

u/may/23

two hydrogen atoms share their valence electrons, both atoms achieve the electronic configuration of noble gas & satisfy the duplet rule.



Types of Covalent Bond:

Depending upon the number of bond pair, a covalent bond is further classified into three types.

- ◊ Single Covalent Bond
- ◊ Double Covalent Bond
- ◊ Triple Covalent Bond

Single Covalent Bond (-)

A covalent bond which is formed by the mutual sharing of one bond pair is called a single covalent bond and it is represented by a single short straight line. The formation of $\text{H}-\text{H}$, $\text{H}-\text{Cl}$, CH_4 are few examples of this type of bonding.

Examples:

Formation of Chlorine Molecule:

One more electron to achieve a stable outer electronic configuration. When two chlorine atoms share their valence

Answer Key For Covalent Bond Concept Map

R Sanford



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