

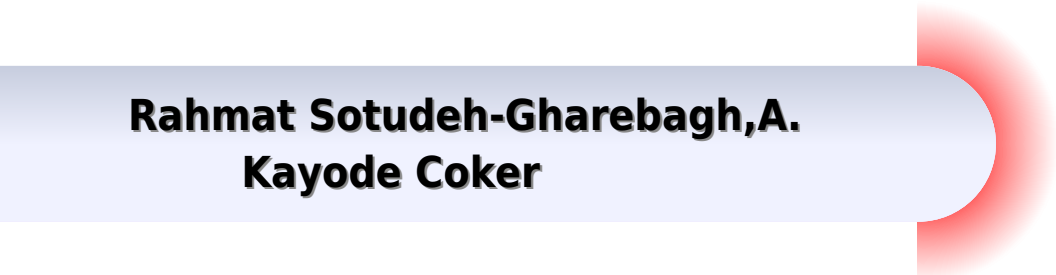
Mole Conversions Worksheet

Solve the following; express your answer using the proper amount of significant figures.

1. How many moles of magnesium are in 3.01×10^{22} atoms of magnesium?
0.0500 moles of magnesium
2. How many molecules are there in 4.00 moles of glucose, $C_6H_{12}O_6$?
 2.41×10^{24} molecules of glucose
3. How many moles are 1.20×10^{25} atoms of phosphorous?
19.9 moles of phosphorous
4. How many moles are in 28 grams of CO_2 ?
0.64 moles of carbon dioxide
5. What is the mass of 5 moles of Fe_2O_3 ?
 8×10^2 grams of iron (III) oxide
6. Find the number of moles of argon in 452 g of argon.
11.3 moles of Argon
7. Find the mass in grams of 2.00×10^{23} molecules of F_2 .
12.6 grams of difluorine
8. Find the mass, in grams, of 1.00×10^{23} molecules of N_2 .
4.65 grams of dinitrogen
9. How many particles are there in 1.43 g of a molecular compound with a molar mass of 233 g? **3.70×10^{21} particles**
10. Aspartame is an artificial sweetener that is 160 times sweeter than sucrose (table sugar) when dissolved in water. The molecular formula of aspartame is $C_{14}H_{18}N_2O_5$.
 - a. Calculate the molar mass of aspartame.
294.34 grams/mole
 - b. How many moles are in 10 g of aspartame?
0.03 moles of aspartame
 - c. How many molecules are in 5 mg of aspartame?
 1×10^{19} molecules of aspartame
 - d. What is the mass, in grams, of 1.56 moles of aspartame?
459 grams of aspartame
 - e. How many atoms of nitrogen are in 1.2 grams of aspartame?
 4.9×10^{21} atoms of nitrogen

Answer Key Mass Mole Conversion Worksheet

**Rahmat Sotudeh-Gharebagh,A.
Kayode Coker**



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Holt Chemistry Ralph Thomas Myers, 2004 **Handbook on Material and Energy Balance Calculations in Material Processing** Arthur E. Morris, Gordon Geiger, H. Alan Fine, 2012-01-03

Lately there has been a renewed push to minimize the waste of materials and energy that accompany the production and processing of various materials. This third edition of this reference emphasizes the fundamental principles of the conservation of mass and energy and their consequences as they relate to materials and energy. New to this edition are numerous worked examples illustrating conventional and novel problem solving techniques in applications such as semiconductor processing, environmental engineering, the production and processing of advanced and exotic materials for aerospace, electronic and structural applications.

Chemical Process Engineering Volume 1 Rahmat Sotudeh-Gharebagh, A. Kayode Coker, 2022-05-03

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Science and Mathematics Lab Manual, 2002-05

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