

Activity A:**Ions**Get the Gizmo ready:

- Click **Reset**.
- Check that sodium and chlorine are still selected.



Introduction: Some of the particles that make up atoms have an electrical charge. Electrons are negatively charged, while protons are positively charged. Particles with opposite charges (+ and –) attract, while particles with the same charge (+ and + or – and –) repel.

Question: What happens when atoms gain or lose electrons?

1. **Count:** Electrons move around the nucleus of atoms in specific shells, shown by the rings around the atoms in the Gizmo. The first ring holds two electrons, and the second holds eight. (If you like, you can hide the inner electrons by selecting **Hide inner electrons**.)

- A. Observe the sodium and chlorine atoms. How many electrons are there total in each atom?

Sodium: 11

Chlorine: 17

- B. Each atom is neutrally charged, which means that each atom has the same number of protons and electrons. Based on this, how many protons are in each atom?

Sodium: 11

Chlorine: 17

2. **Transfer:** Drag electrons from one atom onto another atom's electron shell. How many electrons can the first shell hold? How many electrons can the second shell hold?

Sodium: 2

Chlorine: 8

3. **Electrostatic:** Drag electrons from a shell of +, and watch another shell's charge of +. The sign indicates the charge of an electron, representing the number of electrons from the number of protons. What is the charge from the sodium to the chlorine atom?

- A. What is the charge of each electron? Sodium: -1 Chlorine: -1

Turn on **Show charge** to check. These charged atoms are called **ions**.

- B. Is each ion stable? Explain.

Yes, the first sodium atom electron shell is full and stable.

Yes, the second chlorine atom electron shell is full and stable.

4. **Electrostatic:** What is the net electrical charge the two ions in the diagram have?

Neutral. They have opposite charges.

Answer Key Of Gizmo Additive Colours

**Salvatore Brischetto, Paolo
Maggiore, Carlo Giovanni Ferro**



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