

MASTE 2

Synthesis & Decomposition Reactions

Part 1

Name: 40 points

Per: _____

A. Below the word equation, write the Chemical equation for the reaction.

B. Determine if the reaction is a Synthesis or Decomposition reaction.

1. ^{monatomic} Copper reacts with ^{diatomic} Chlorine gas to produce ^{ionic} Copper (II) chloride.



+, Synthesis

2. Ammonia (NH_3) reacts with Hydrochloric acid (HCl) to produce ^{ionic} Ammonium chloride.



+, Synthesis

3. Aluminum Hydroxide produces Aluminum oxide and Water.



+, Decomposition

4. ^{metallic} Sodium metal reacts with ^{diatomic} Oxygen gas to produce ^{ionic} Sodium oxide.



+, Synthesis

5. ^{ionic} Copper (II) oxide produces ^{metallic} Copper metal and ^{diatomic} Oxygen gas.



-, Decomposition

Answer Key To Decomposition Synthesis Reactions

CL Gary



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