

Acids and bases review sheet Answer key

1. List the differences between an acid and a base. Your description should include such things as whether both are electrolytes, which contains a hydrogen ion and which includes a hydroxide ion, and you should also include two examples of an acid and two examples of a base that are present in your household.

Acid	base
Conducts electricity	conducts electricity
Contains hydrogen ion	contains hydroxide ion
Proton donor	proton acceptor
Turns litmus paper red	turns litmus paper blue
React with most metals	releases hydroxide ion
Releases hydronium	Ex: Windex,
Ex: lemon	bleach
Vinegar	soap
Battery acid (sulfuric acid)	

2. Listed below are some of the properties of acids and bases. Fill in the blanks with the appropriate word, **acids** or **bases**:

- A produce hydrogen ions (H^+) in solution
- B have a bitter taste
- B produce hydroxide ions (OH^-) ions
- A react with bases to form salts
- A have a sour taste
- B have a slippery, soapy feel
- B react with acids to form salts
- A react with many metals

3. Write an equation for the dissociation of the following **acids** in water: follow example

- $HClO_4 \rightarrow H^+ + ClO_4^-$
- $H_2SO_4 \rightarrow 2H^+ + SO_4^{2-}$
- $HC_2H_3O_2 \rightarrow H^+ + C_2H_3O_2^-$
- $H_2S \rightarrow 2H^+ + S^{2-}$
- $HCl \rightarrow H^+ + Cl^-$
- $HNO_3 \rightarrow H^+ + NO_3^-$

4. Write an equation for the dissociation (ionization) of the following **bases** in water: (follow example)

- $Mg(OH)_2 \rightarrow Mg^{2+} + 2OH^-$
- $NaOH \rightarrow Na^+ + OH^-$
- $Ba(OH)_2 \rightarrow Ba^{2+} + 2(OH^-)$
- $KOH \rightarrow K^+ + OH^-$
- $Mg(OH)_2 \rightarrow Mg^{2+} + 2(OH^-)$
- $LiOH \rightarrow Li^+ + OH^-$

5) A. What type of reaction is a **neutralization** reaction? **Double replacement**

B. Write the net equation for neutralization:



Answer Key To Describing Acids And Bases

**Peter J. Mikulecky, Michelle Rose
Gilman, Kate Brutlag**



Answer Key To Describing Acids And Bases:

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PHARMACEUTICAL ANALYSIS Mrs. Sandhya. Kota,Ms. Araf Mahefuzabibi H ,Dr. Rohit Singh ,Dr. S. Kambhoja ,Dr. Priti Rajendra Kale , The past several decades have seen a tremendous change in the area of pharmaceutical analysis mostly due to the growing complexity of the difficulties we seek to solve the integration of interdisciplinary methods and the development of analytical techniques This book Pharmaceutical Analysis aims to provide a thorough review that takes into account the complex character of the state of the art studies in this area It is intended to provide experts researchers and students with the fundamental abilities and information required to successfully and properly traverse the challenging field of pharmaceutical analysis The effort was motivated by the understanding that thorough analysis procedures are essential to generating accurate and significant information In a time where data is easily accessible and plentiful the difficulty is not just gathering information but also arranging analyzing and assessing it critically in order to make significant judgments By giving readers a comprehensive grasp of both qualitative and quantitative analytical methodologies within the context of pharmaceutical analysis this book seeks to close the knowledge gap between theoretical principles and actual execution Each chapter has been painstakingly written to cover a broad spectrum of subjects ranging from the sophisticated techniques used in pharmaceutical analysis to the basic concepts of analytical chemistry Current concerns include the use of cutting edge technology in drug research the morality of data collecting and the value of multidisciplinary methods have received special attention This book shows how to use a variety of analytical approaches to different parts of pharmaceutical analysis via case studies and real world examples Putting this book together has been a joyful and difficult journey It has included a thorough analysis of the literature professional contacts and the synthesis of many viewpoints Through the joint efforts of researchers academics and practitioners the information has been molded and made relevant to both present and future research pursuits

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Science Explorer Physical Science Michael J. Padilla, Ioannis Miaculis, Martha Cyr, **Paediatric Intensive Care** Peter Barry, Kevin Morris, Tariq Ali, 2017 One in 70 children are admitted to paediatric intensive care PIC at some time during childhood Most paediatric junior doctors will rotate through PIC and will be involved in organising acute intensive care for critically ill children The range of children and their illnesses going through PIC is vast making it a hugely diverse specialty A critically ill child will end up there regardless of their underlying disease and as a result consultants in PIC must be true generalists and need to acquire knowledge and skills in all areas of paediatrics as well as acquiring significant knowledge of anaesthesia and surgery From setting up the ventilator to managing low cardiac output Paediatric Intensive Care gives practical and realistic advice for children's doctors and nurses in intensive care Information is presented in easily accessible 5 minute chunks to enable you to quickly get the answers you

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Respiratory Care: Patient Assessment and Care Plan Development David C. Shelledey,Jay I. Peters,2021-02-08

Respiratory Care Patient Assessment and Care Plan Development Second Edition describes the purpose of patient

assessment and then guides the reader through the process of reviewing existing data in the medical record *Comparative Aspects of Extracellular Acid-Base Balance* Jean-Paul Truchot, 2012-12-06 Interest in comparative acid base physiology has considerably grown during last decades even in the absence of major technical or conceptual advances This is firstly because it has become clear that the extracellular acid base state reflects the performance of many exchange functions at the organism level respiration and ventilation of the gas exchange surfaces metabolism and osmoregulation Such functions are much influenced by ambient conditions and the measurement of acid base parameters thus provides useful information about the organism's responses to environmental challenges Secondly many processes at the molecular level are now known to be pH sensitive and acid base regulation thus appears to be a major requirement for the functional integrity of cells and organisms How extracellular acid base balance can be maintained in a wide variety of animals living in different conditions is the subject of this book The approach is comparative and environmental throughout All body fluids share similar buffer properties and common physicochemical principles apply to any acid base system However in accord with differing designs and constraints along animal evolution varying effector organs and mechanisms are at work to maintain an appropriate acid base state in the organism Particular emphasis is placed on the fundamental differences between water and air breathers and on the acid base and respiratory problems arising at the transition from an aquatic to a terrestrial life Also the complex array of factors influencing the acid base state in water dwelling animals is thoroughly discussed **Nephrology, An Issue of Critical Care Clinics** John Kellum, 2016-01-07 Dr John Kellum has assembled an essential update on the topic of Nephrology as it relates to Critical Care Medicine Articles include Diagnostic criteria Biomarkers for AKI Sepsis induced AKI Drug induced AKI Cardio renal syndrome Surgery Associated AKI Contrast induced AKI Principles of Fluid Therapy Fluid composition and clinical effects Renal replacement therapy and Understanding acid base **Cambridge IGCSE & O Level Chemistry: Exam Success** Lawrie Ryan, Roger Norris, 2021-09-30 The Cambridge IGCSE O Level Chemistry Exam Success Guide fully supports the latest Cambridge IGCSE 0620 O Level 5070 syllabuses and is suitable for use alongside our Complete and Essential IGCSE Chemistry series The Guide helps students cope with the increased rigour of linear IGCSEs by bringing clarity and focus to exam preparation and by providing explicit exam guidance Learners can recap content through easy to digest chunks apply this via targeted revision activities review and reflect on their work and use exam practice and worked examples to achieve best results The Chemistry Exam Success Guide is written by Roger Norris Cambridge chief examiner for IGCSE Chemistry and Lawrie Ryan series editor of the Essential Science series Students can benefit from their expertise and excellent understanding of what support learners need in order to reach their full potential Other resources are also available a Practical Workbook Student Books and Workbooks The Practical Workbook helps students to achieve practical exam success The Complete or Essential Student Book is at the heart of delivering the course and is available in print online or in a great value print and online pack The Complete or Essential Workbook is for independent practice and

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