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ppm Answer- (grams analyte/grams sample)x10^6

Molarity Answer- moles analyte/liter of solution

Volume Percent Answer- (volume solute/volume soution)x100

Volume ppm Answer- (volume solute/volume solution)x10^6

kilo- Answer- 10<sup>3</sup>

deci- Answer- 10^-1

centi- Answer- 10^-2

milli- Answer- 10^-3

micro- Answer- 10^-6

nano- Answer- 10^-9

pico- Answer- 10^-12

femto- Answer- 10^-15

weight percent. Answer- (grams analyte/grams sample)x100

ppt Answer- (grams analyte/grams sample)x10<sup>a</sup>3

ppt simplified Answer- gram analyte/liter solution

# **Analytical Chemistry Exam Study Guide**

**Sabine Zange** 

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