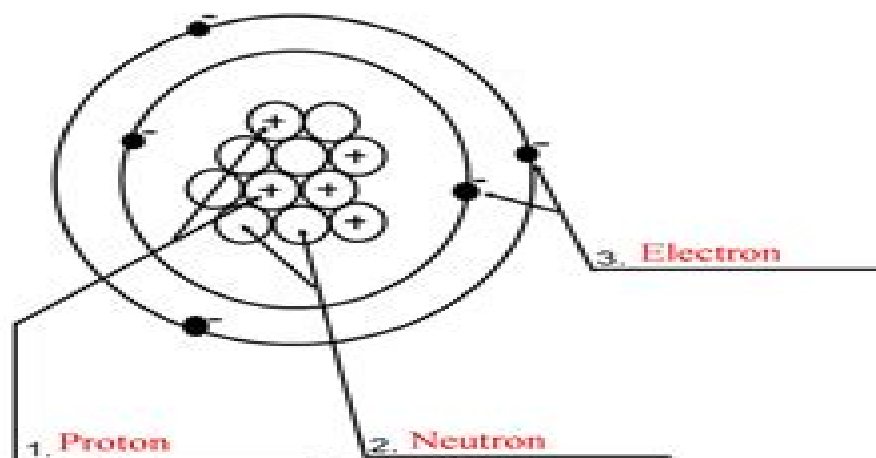


Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?

Positive (+1)

5. What type of charge does a neutron have?

Neutral (0)

6. What type of charge does an electron have?

Negative (-1)

7. Which two subatomic particles are located in the nucleus of an atom?

Protons and neutrons

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?

35

9. What is the atomic number of the atom in the diagram above?

5

10. What is the atomic mass/mass number of the atom in the diagram above?

$(11+11=22)$ *protons and neutrons added together) 22 is the atomic mass/mass number

11. How many protons are in the nucleus of an atom with an atomic number of 15?

15 *number of protons is the same as the atomic number

12. How many electrons are in the nucleus of an atom with an atomic number of 20?

20

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?
(use Periodic Table for mass)

30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25..... $55-25=30$

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?

Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?

Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

Analyzing Atomic Structure Worksheet Answers

DJ Losen



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Table of Contents Analyzing Atomic Structure Worksheet Answers

1. Understanding the eBook Analyzing Atomic Structure Worksheet Answers
 - The Rise of Digital Reading Analyzing Atomic Structure Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Analyzing Atomic Structure Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Analyzing Atomic Structure Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Analyzing Atomic Structure Worksheet Answers
 - Personalized Recommendations
 - Analyzing Atomic Structure Worksheet Answers User Reviews and Ratings

- Analyzing Atomic Structure Worksheet Answers and Bestseller Lists
- 5. Accessing Analyzing Atomic Structure Worksheet Answers Free and Paid eBooks
 - Analyzing Atomic Structure Worksheet Answers Public Domain eBooks
 - Analyzing Atomic Structure Worksheet Answers eBook Subscription Services
 - Analyzing Atomic Structure Worksheet Answers Budget-Friendly Options
- 6. Navigating Analyzing Atomic Structure Worksheet Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Analyzing Atomic Structure Worksheet Answers Compatibility with Devices
 - Analyzing Atomic Structure Worksheet Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Analyzing Atomic Structure Worksheet Answers
 - Highlighting and Note-Taking Analyzing Atomic Structure Worksheet Answers
 - Interactive Elements Analyzing Atomic Structure Worksheet Answers
- 8. Staying Engaged with Analyzing Atomic Structure Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Analyzing Atomic Structure Worksheet Answers
- 9. Balancing eBooks and Physical Books Analyzing Atomic Structure Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Analyzing Atomic Structure Worksheet Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Analyzing Atomic Structure Worksheet Answers
 - Setting Reading Goals Analyzing Atomic Structure Worksheet Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Analyzing Atomic Structure Worksheet Answers
 - Fact-Checking eBook Content of Analyzing Atomic Structure Worksheet Answers
 - Distinguishing Credible Sources

-
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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