

ite:	

# COUNTING ATOMS

<ol> <li>Chemical:</li> </ol>	formulas can:	have three	components.
-------------------------------	---------------	------------	-------------

a):	The	which are represented by a single uppercase letter, or represented by an
	uppercase letter follow	d by a lowercase letter. This tells you the types of elements in the compound.

- b) The \_\_\_\_\_\_ which are numbers that are found on the lower \_\_\_\_\_ hand side of each element symbol. This tells you the number of \_\_\_\_\_ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only atom of this element in the molecule.
- c) The \_\_\_\_\_ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be \_\_\_\_\_ by the \_\_\_\_\_ on the outside of the bracket found on the lower \_\_\_\_\_ hand side of that bracket.
- 2. Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

I) NaOH	
Element	# of Atoms
Total	
4) Li <sub>2</sub> SO <sub>4</sub>	70
Element	# of Atoms
Total	
7) H,COCH,	
Element	# of Atoms
Total	
10) NH,C,H,4	O <sub>2</sub>
Element	# of Atoms
Total	
13) Ca <sub>2</sub> (PO <sub>4</sub> )	
Element	# of Atoms
WALLES THE STATE OF THE STATE O	n or receive

Total

# of Atoms
# of Atoms

Element	# of Atoms
7.4	
Total	
8) Mg(OH) <sub>2</sub>	
Element	M of Atoms

Element	# of Atoms
Total	

11) NaC <sub>2</sub> H <sub>3</sub> O	
Element	# of Atoms
Total	

14) Fe <sub>2</sub> (CO <sub>3</sub> ),	122
Element	# of Atoms
Total	
Lotal	

Element	# of Atoms
	9

6) CaClO,	
Element	# of Atoms
	3
Total	

9) AI(OH),	
Element	# of Atoms
	5
Total	

12) (NH <sub>4</sub> ),PO <sub>4</sub>	
Element	# of Atoms
	3
Total	

15) Al <sub>2</sub> (SO <sub>3</sub> ),	123
Element	# of Atoms
Total	

# **Answer Key For Counting Atoms In Compounds**

**D. Michael P. Mingos** 

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