Name:	Plantan.	Class period:
	Date:	Colors Deringer.

#### Unit: Nuclear Chemistry Test Review



#### What to study?

- Quizzes: #1
- Homework: worksheets #1-#4
- Notes: Properties of Radiation → Applications of Nuclear Chemistry
- Bell Ringers: Matching, Half-lives, Writing Transmutations

#### What will the questions be like? What can I expect?

This test will be just like the previous quizzes you have taken.

Matching

Short answer

Multiple choice

Calculations

#### Sample questions/Important Topics

- 1. Fill in the best answer for each of the following:

  - b) Three types of radiation include \_\_\_\_\_\_, \_\_\_\_\_, and
  - c) During radioactive decay, the \_\_\_\_\_\_ will emit particles of radiation in order to become more stable.
  - d) The reactions inside the sun is an example of a \_\_\_\_\_\_ reaction.
  - e) If you wanted to block \_\_\_\_\_\_ radiation, paper or clothing can be used.
  - Radioactive isotopes, or \_\_\_\_\_\_\_, have different mass numbers but the same atomic number and will decay over time.
  - g) \_\_\_\_\_\_ is used to calculate the date of a specimen that is embedded deep within the earth.
  - is the time it takes for half a sample of nuclei to decay into more stable nuclei.
  - i) Fusion reactions release \_\_\_\_\_\_ energy than fission reactions.
  - For elements that have an atomic number less than 20, stable nuclei have a neutron:proton ratio equal to \_\_\_\_\_\_\_.
  - k) \_\_\_\_\_\_ is the spontaneous emission of radiation by an unstable nucleus.
  - A common isotope used to diagnose thyroid disease is called
  - m) During \_\_\_\_\_\_ decay, the atomic number will increase by 1 and the mass number will remain unchanged.
  - n) \_\_\_\_\_ used alpha particles in his Gold Foil Experiment.

# **Answer Guide To Nuclear Chemistry**

John T. Stimus

## **Answer Guide To Nuclear Chemistry:**

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