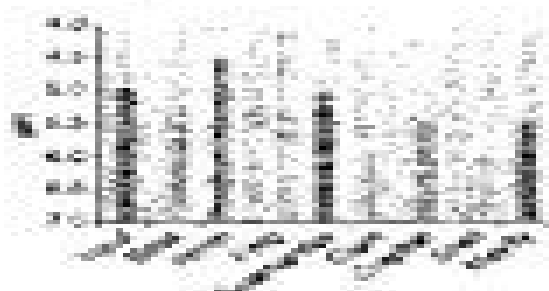


8.

Examine Figure 4.7 and then answer the questions using the graph below:



- At what pH range can all selected species survive in this acidic, basic, or neutral? $pH=7$ Neutral to slightly acidic
- At what pH range do the most species survive? Is this acidic, basic, or neutral? $pH=7$ Acidic
- Which species are the most sensitive to acidic environments? How can you tell? $\text{Amphibians \& Insects are most sensitive, because they are the first species to become not tolerant to increasing acidity.}$

9.

The following are chemical reactions for the combustion of various fuels.

Balance the equations with their proper stoichiometric coefficients. Be sure to show

all your work.

- Methane combustion: $CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O$
- Propane combustion: $C_3H_8 + 5O_2 \rightarrow 3CO_2 + 4H_2O$
- Ethanol combustion: $C_2H_5OH + 3O_2 \rightarrow 2CO_2 + 3H_2O$
- Gasoline combustion: $2C_8H_{18} + 25O_2 \rightarrow 16CO_2 + 18H_2O$

Answer Key For Envriionmenal Science Friedland

JA Banks



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