

Name: Answer Key Date: \_\_\_\_\_ Per.: \_\_\_\_\_

# Covalent Bond Practice

1. Fill in the missing information on the chart.

| Element    | # of Protons | # of Electrons | # of Valence Electrons | # of electrons to fill outer shell. |
|------------|--------------|----------------|------------------------|-------------------------------------|
| Carbon     | 6            | 6              | 4                      | 4                                   |
| Hydrogen   | 1            | 1              | 1                      | 1                                   |
| Chlorine   | 17           | 17             | 7                      | 1                                   |
| Helium     | 2            | 2              | 2                      | 0                                   |
| Phosphorus | 15           | 15             | 5                      | 3                                   |
| Oxygen     | 8            | 8              | 6                      | 2                                   |
| Sulfur     | 16           | 16             | 6                      | 2                                   |
| Nitrogen   | 7            | 7              | 5                      | 3                                   |

2. For each of the following covalent bonds:

- Write the symbols for each element.
- Draw a Lewis Dot structure for the valence shell of each element.
- Rearrange the electrons to pair up electrons from each atom.
- Draw circles to show the sharing of electrons between each pair of atoms.
- Draw the bond structure using chemical symbols and lines. Use one line for each pair of electrons that is shared.
- Write the chemical formula for each molecule.

a) Hydrogen + Hydrogen



b) Chlorine + Chlorine



c) Hydrogen + Chlorine



d) Hydrogen + Oxygen



e) Nitrogen + Hydrogen

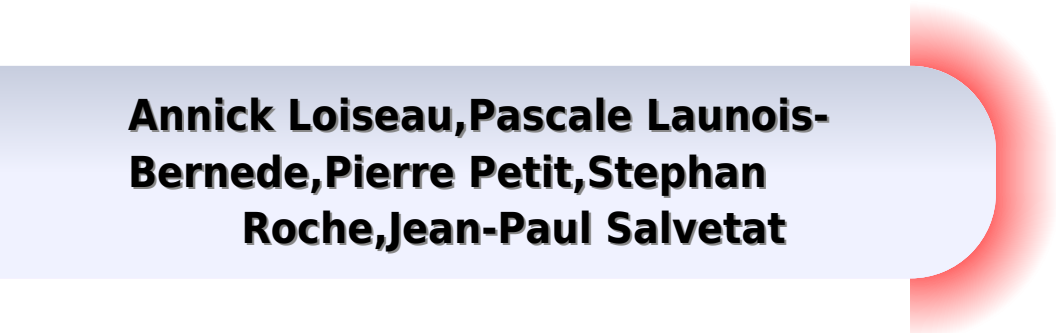


f) Carbon + Hydrogen



# Answer Key To Bonding Basics Covalent Bonds

**Annick Loiseau, Pascale Launois-  
Bernede, Pierre Petit, Stephan  
Roche, Jean-Paul Salvetat**



## **Answer Key To Bonding Basics Covalent Bonds:**

**Study Guide to Accompany Basics for Chemistry** Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students *Fundamentals of Chemistry (English Edition)* Dr. Rubby Mishra,,Dr. Krishna Kumar Singh ,2021-02-01 Buy Latest Fundamentals of Chemistry B Sc 1 Sem Chemistry Book especially designed for U P State universities by Thakur Publication *An Introduction to Chemistry* Michael Mosher,Paul Kelter,2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry *Basics for Chemistry* David A. Ucko,2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of

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with this easy to follow textbook Analytical chemistry is the study of chemical composition concerned with analyzing materials to discover their constituent substances the amounts in which these substances are present and more Since materials exist in different states and undergo reactions analytical chemistry is also concerned with chemical equilibria the state at which various reactants and substances will undergo no observable chemical change without outside stimulus This field has an immense range of practical applications in both industry and research and is a highly desirable area of expertise for the next generation of chemists Basics of Analytical Chemistry and Chemical Equilibria provides an introduction to this foundational subject ideal for specialized courses It introduces not only the core concepts of analytical chemistry but cultivates mastery of various instrumental methods by which students and researchers can undertake their own analyses Now updated to include the latest research and expanded coverage Basics of Analytical Chemistry and Chemical Equilibria promises to situate a new generation of readers in this growing field Readers of the second edition of Basics of Analytical Chemistry and Chemical Equilibria will also find A new chapter on structure determination Revised and expanded descriptions of chemical instrumentation You try it exercises throughout to further develop practical student knowledge Companion website of associated materials including end of chapter solutions spreadsheets for student use and more Basics of Analytical Chemistry and Chemical Equilibria is an ideal textbook for students in chemistry biochemistry and environmental science as well as students in related fields including chemical engineering and materials science for whom analytical chemistry offers a useful toolset

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*Understanding Carbon Nanotubes* Annick Loiseau, Pascale Launois-Bernede, Pierre Petit, Stephan Roche, Jean-Paul Salvetat, 2006-07-07 This volume presents the foundations of carbon nanotube science reviewing recent developments and prospects for practical application Each chapter summarizes relevant concepts from physics chemistry or materials science followed by detailed reports on topics including polymorphism and mircostructure of carbon synthesis and growth structural analysis by electron microscopy spectroscopic methods electronic structure transport mechanical and surface properties of nanotubes and composites

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Foundation Engineering Handbook Hsai-Yang Fang, 2013-06-29 More than ten years have passed since the first edition was published During that period there have been a substantial number of changes in geotechnical engineering especially in

the applications of foundation engineering As the world population increases more land is needed and many soil deposits previously deemed unsuitable for residential housing or other construction projects are now being used Such areas include problematic soil regions mining subsidence areas and sanitary landfills To overcome the problems associated with these natural or man made soil deposits new and improved methods of analysis design and implementation are needed in foundation construction As society develops and living standards rise tall buildings transportation facilities and industrial complexes are increasingly being built Because of the heavy design loads and the complicated environments the traditional design concepts construction materials methods and equipment also need improvement Further recent energy and material shortages have caused additional burdens on the engineering profession and brought about the need to seek alternative or cost saving methods for foundation design and construction

### **Hazardous Materials Chemistry for Emergency**

**Responders** Robert Burke,2002-11-25 The second edition of a bestseller Hazardous Materials Chemistry for Emergency Responders continues to provide the fundamentals of street chemistry required by emergency response personnel The information presented will assist you in responding to specific chemical spills including identifying the exact chemicals involved and their individual

**Introduction to Biological Imaging** Manfred Auer,2024-04-29 Discover what biological imaging is able to accomplish in this up to date textbook One of the fundamental goals of biology is to understand how living organisms establish and maintain their spatiotemporal organization of the biochemical cell biological and developmental biology processes that sustain life Biological systems are inherently complex with a large number of components needed to sustain cellular function In order to understand any complex system one must determine its composition by identifying the components it is made of how each of these components function and carry out their specific task and how they interact with one another to function together To grasp the link of such changes to physiological cell and tissue function and or pathogenesis disease progression we need to understand how modifications alter macromolecular function macromolecular interactions and or spatiotemporal distribution and overall supramolecular structural organization Biological imaging holds the key to understanding spatiotemporal organization and will thus be increasingly important for the next generations of biological and biochemical researchers Introduction to Biological Imaging provides the first comprehensive textbook surveying this subject It elucidates the fundamental principles underlying the capture and production of bioimages the requirements of image analysis and interpretation and some key problems and solutions in bioimaging It includes everything experimental biologists need to incorporate appropriate bioimaging solutions into their work Introduction to Biological Imaging readers will also find Coverage of all major types of biological imaging including medical imaging cellular imaging macromolecular imaging and more Advice on preparing samples for various imaging methods Specific examples in each chapter connecting bioimaging process to the production of real experimental data Introduction to Biological Imaging is a valuable introduction for undergraduate or graduate students in courses relating to bioimaging as well as scientists and

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