

## Naming Acids and Bases Answers

Name the following acids and bases:

- |    |                                |                             |
|----|--------------------------------|-----------------------------|
| 1) | NaOH                           | <u>sodium hydroxide</u>     |
| 2) | H <sub>2</sub> SO <sub>3</sub> | <u>sulfurous acid</u>       |
| 3) | H <sub>2</sub> S               | <u>hydrosulfuric acid</u>   |
| 4) | H <sub>3</sub> PO <sub>4</sub> | <u>phosphoric acid</u>      |
| 5) | NH <sub>3</sub>                | <u>ammonia</u>              |
| 6) | HCN                            | <u>hydrocyanic acid</u>     |
| 7) | Ca(OH) <sub>2</sub>            | <u>calcium hydroxide</u>    |
| 8) | Fe(OH) <sub>3</sub>            | <u>iron (III) hydroxide</u> |
| 9) | H <sub>3</sub> P               | <u>hydrophosphoric acid</u> |

Write the formulas of the following acids and bases:

- |     |                       |                                    |
|-----|-----------------------|------------------------------------|
| 10) | hydrofluoric acid     | <u>HF</u>                          |
| 11) | hydroselenic acid     | <u>H<sub>2</sub>Se</u>             |
| 12) | carbonic acid         | <u>H<sub>2</sub>CO<sub>3</sub></u> |
| 13) | lithium hydroxide     | <u>LiOH</u>                        |
| 14) | nitrous acid          | <u>HNO<sub>2</sub></u>             |
| 15) | cobalt (II) hydroxide | <u>Co(OH)<sub>2</sub></u>          |
| 16) | sulfuric acid         | <u>H<sub>2</sub>SO<sub>4</sub></u> |
| 17) | beryllium hydroxide   | <u>Be(OH)<sub>2</sub></u>          |
| 18) | hydrobromic acid      | <u>HBr</u>                         |

# Answers For Naming Acids

**Neil D. Jespersen, Alison Hyslop**



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