

Chemistry - Section 3: Equations, Calculations and Electrolysis QUESTIONS AND ANSWERS

1. (aq) - ANSWER-aqueous (dissolved in water)
2. (g) - ANSWER-gas
3. (l) - ANSWER-drinks
4. (s) - ANSWER-strong
5. 2H_2 equals - ANSWER-there are already 2 hydrogen atoms in H_2 . Therefore while a 2 is positioned at the back of it, it equals to 4 hydrogen atoms
6. a 0% yield way - ANSWER-no reactants have been converted into products
7. a 100% yield means - ANSWER-to procure all of the product you anticipated to
8. a strong salt containing water of crystallization is - ANSWER-hydrated
9. Ar - ANSWER-relative atomic mass
10. carbon has an Ar of 12 - ANSWER-consequently one mole of carbon weighs precisely 12g
11. definition approximately moles and mass - ANSWER-one mole of atoms or molecules of any substance will have a mass of grams identical to the relative particle mass (Ar or Mr) for that substance
12. empirical formulae using discount - ANSWER-1. Place a rubber bung right into a take a look at tube with a small hollow in the long run and weigh them using a mass stability.
 2. Take bung out and spread out a small amount of copper(II) oxide inside the center of the tube
 3. Re-insert bung and weigh the tube once more. Set up apparatus.
 4. Expel air from the test tube by using lightly turning at the gas. After five seconds, mild the gasoline with the aid of conserving a burning splint next to the hole in the long run of the test tube.
 5. Use a Bunsen burner to warmth the copper(II) oxide for 10 minutes, or until solid adjustments from black to brownish-red color.
 6. Turn off Bunsen burner and go away the check tube to chill.
 7. Turn off gas and weigh the test tube with the bung and contents.
13. Formula for variety of moles - ANSWER-mass / Mr
14. how are you going to locate the empirical system using combustion? - ANSWER-1. Heat a crucible until it's miles pink hot. This will ensure it's far smooth without a traces of oil or water.
 2. Leave the crucible to cool, then weigh it, along side its lid.

Answers To Chemistry Section Assessme

Leo J. Malone, Theodore O. Dolter



Answers To Chemistry Section Assessme:

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