Chemthink guided notes - Ions

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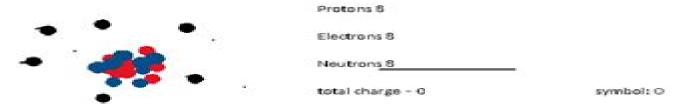
At certain times atoms can gain or lose one or more ____electrons_____. When this happens we say that the atom has formed an _ion_____. Ions are just individual atoms or groups of atoms that has a charge. This charge comes from an imbalance of in the number of ___positive_____ protons and __negative_____ electrons.

Since the electrons don't contribute to the ____mass_____ of the atom, gaining or losing electrons does not change the overall mass of the atom. It does affect it in other ways, lons can behave very differently than the atoms from which they were formed. As we will see in later units, forming ions is one way that atoms bond together to make new substances.

Draw a picture of the lithium atom and write the particles in the particle inventory

9	25	Protons 3	
		Electrons 3	
8		Neutrons 4	<u></u>
-	9	total charge - O	symbol: Li
What happens when an electron is lost?			symbol?

Draw a picture of the oxygen atom and write the particles in the particle inventory



What happens when two electrons are gained?

Answers To Chemthink Questions Ions

Buchanan

Answers To Chemthink Questions Ions:

Answers to Revision Questions for Higher Chemistry Buchanan, 2000-12-01

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